

Introduction to Synchrotron Radiation and it's Applications

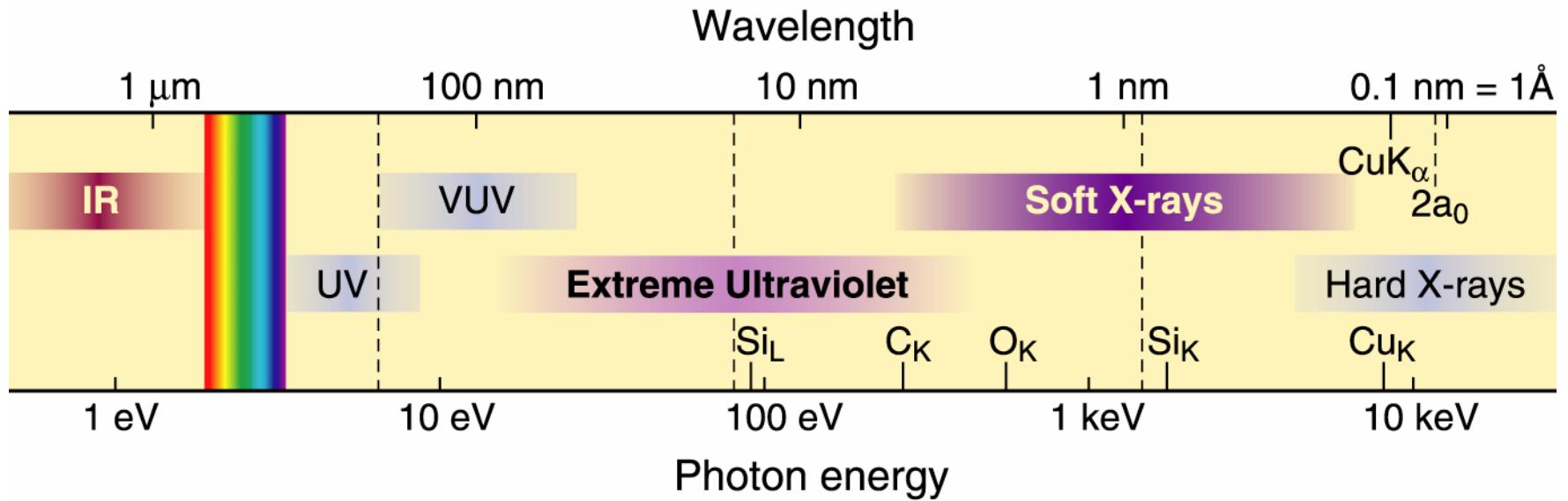
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Lawrence Berkeley National Laboratory**

(<http://www.coe.berkeley.edu/AST/sxreuv>)



The Short Wavelength Region of the Electromagnetic Spectrum

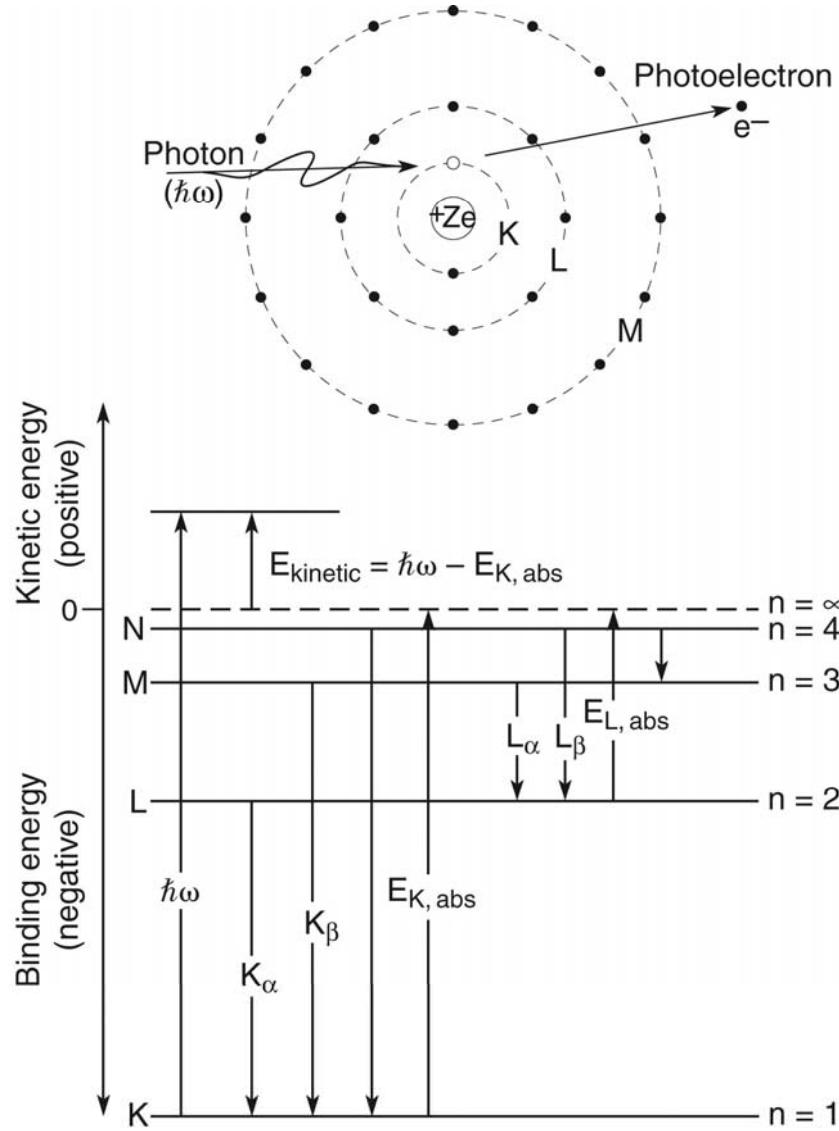


- See smaller features
- Write smaller patterns
- Elemental and chemical sensitivity

$$\hbar\omega \cdot \lambda = hc = 1239.842 \text{ eV nm} \quad (1.1)$$



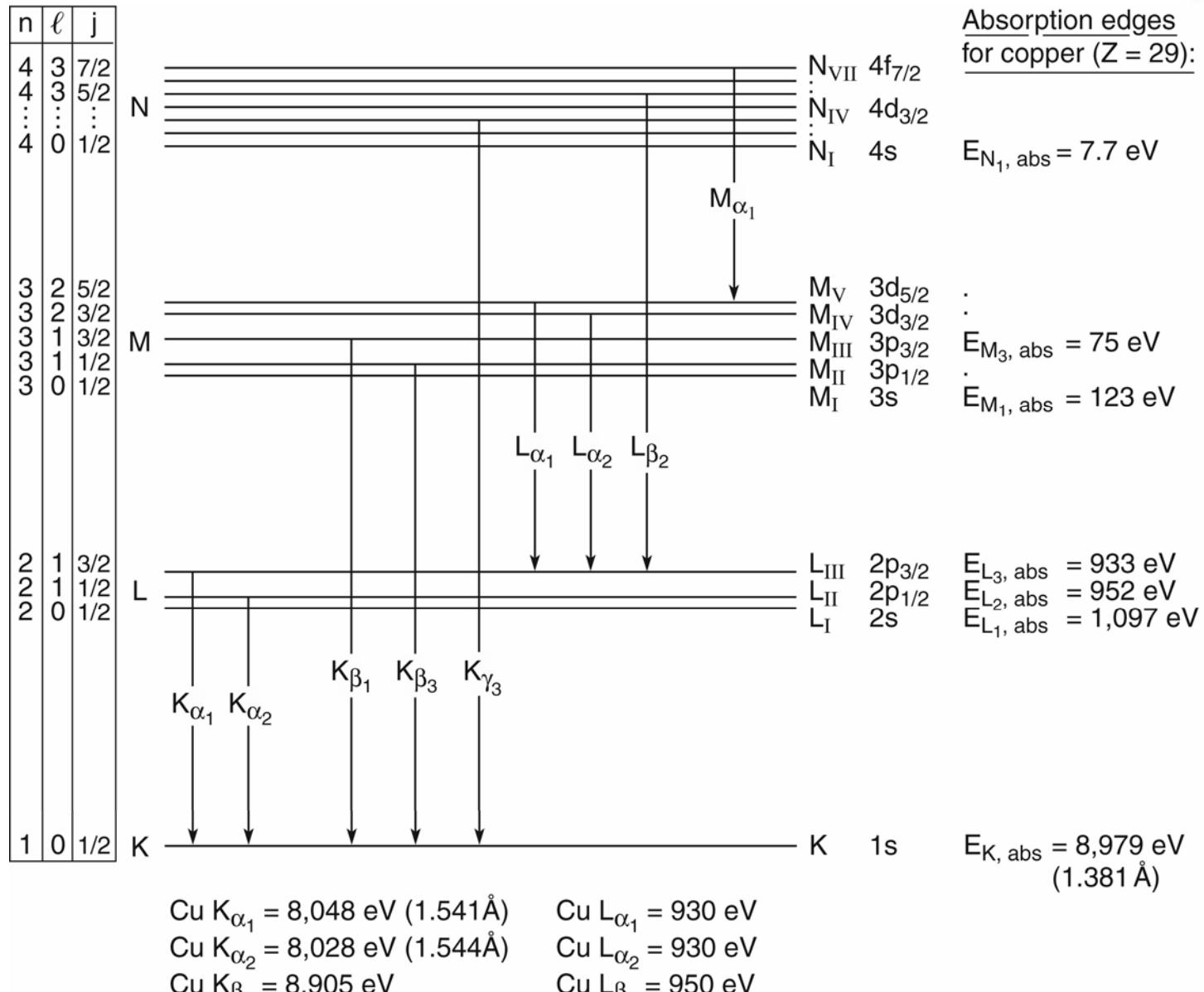
Characteristic Absorption Edges for Almost all Elements in this Spectral Region



Element	Z	K_{abs} -edge (eV)	L_{abs} -edge (eV)
Be	4	112	—
C	6	284	—
N	7	410	—
O	8	543	—
Al	13	1,560	73
Si	14	1,839	99
S	16	2,472	163
Ca	20	4,039	346
Ti	22	4,966	454
V	23	5,465	512
Cr	24	5,989	574
Fe	26	7,112	707
Ni	28	8,333	853
Cu	29	8,979	933
Se	34	12,658	1,434
Mo	42	20,000	2,520
Sn	50	29,200	3,929
Xe	54	34,561	4,782
Pt	78	78,395	11,564
Au	79	80,725	11,919



Energy Levels, Quantum Numbers, and Allowed Transitions for the copper Atom





Electron Binding Energies, in Electron Volts (eV), for the Elements in their Natural Forms



Element	K 1s	L ₁ 2s	L ₂ 2p _{1/2}	L ₃ 2p _{3/2}	M ₁ 3s	M ₂ 3p _{1/2}	M ₃ 3p _{3/2}	M ₄ 3d _{3/2}	M ₅ 3d _{5/2}	N ₁ 4s	N ₂ 4p _{1/2}	N ₃ 4p _{3/2}
1 H	13.6											
2 He	24.6 ^b											
3 Li	54.7 ^b											
4 Be	111.5 ^b											
5 B	188 ^b											
6 C	284.2 ^b											
7 N	409.9 ^b	37.3 ^b										
8 O	543.1 ^b	41.6 ^b										
9 F	696.7 ^b											
10 Ne	870.2 ^b	48.5 ^b	21.7 ^b	21.6 ^b								
11 Na	1070.8 ^c	63.5 ^c	30.4 ^c	30.5 ^b								
12 Mg	1303.0 ^c	88.6 ^b	49.6 ^c	49.2 ^c								
13 Al	1559.6	117.8 ^b	72.9 ^b	72.5 ^b								
14 Si	1838.9	149.7 ^b	99.8 ^b	99.2 ^b								
15 P	2145.5	189 ^b	136 ^b	135 ^b								
16 S	2472	230.9 ^b	163.6 ^b	162.5 ^b								
17 Cl	2822.4	270.2 ^b	202 ^b	200 ^b								
18 Ar	3205.9 ^b	326.3 ^b	250.6 ^b	248.4 ^b	29.3 ^b	15.9 ^b	15.7 ^b					
19 K	3608.4 ^b	378.6 ^b	297.3 ^b	294.6 ^b	34.8 ^b	18.3 ^b	18.3 ^b					
20 Ca	4038.5 ^b	438.4 ^c	349.7 ^c	346.2 ^c	44.3 ^c	25.4 ^c	25.4 ^c					
21 Sc	4492.8	498.0 ^b	403.6 ^b	398.7 ^b	51.1 ^b	28.3 ^b	28.3 ^b					
22 Ti	4966.4	560.9 ^c	461.2 ^c	453.8 ^c	58.7 ^c	32.6 ^c	32.6 ^c					
23 V	5465.1	626.7 ^c	519.8 ^c	512.1 ^c	66.3 ^c	37.2 ^c	37.2 ^c					
24 Cr	5989.2	695.7 ^c	583.8 ^c	574.1 ^c	74.1 ^c	42.2 ^c	42.2 ^c					
25 Mn	6539.0	769.1 ^c	649.9 ^c	638.7 ^c	82.3 ^c	47.2 ^c	47.2 ^c					
26 Fe	7112.0	844.6 ^c	719.9 ^c	706.8 ^c	91.3 ^c	52.7 ^c	52.7 ^c					
27 Co	7708.9	925.1 ^c	793.3 ^c	778.1 ^c	101.0 ^c	58.9 ^c	58.9 ^c					
28 Ni	8332.8	1008.6 ^c	870.0 ^c	852.7 ^c	110.8 ^c	68.0 ^c	66.2 ^c					
29 Cu	8978.9	1096.7 ^c	952.3 ^c	932.5 ^c	122.5 ^c	77.3 ^c	75.1 ^c					
30 Zn	9658.6	1196.2 ^b	1044.9 ^b	1021.8 ^b	139.8 ^b	91.4 ^b	88.6 ^b	10.2 ^b	10.1 ^b			
31 Ga	10367.1	1299.0 ^b	1143.2 ^c	1116.4 ^c	159.5 ^c	103.5 ^c	103.5 ^c	18.7 ^c	18.7 ^c			
32 Ge	11103.1	1414.6 ^b	1248.1 ^b	1217.0 ^b	180.1 ^b	124.9 ^b	120.8 ^b	29.0 ^b	29.0 ^b			
33 As	11866.7	1527.0 ^b	1359.1 ^b	1323.6 ^b	204.7 ^b	146.2 ^b	141.2 ^b	41.7 ^b	41.7 ^b			
34 Se	12657.8	1652.0 ^b	1474.3 ^b	1433.9 ^b	229.6 ^b	166.5 ^b	160.7 ^b	55.5 ^b	54.6 ^b			
35 Br	13473.7	1782.0 ^b	1596.0 ^b	1549.9 ^b	257 ^b	189 ^b	182 ^b	70 ^b	69 ^b			
36 Kr	14325.6	1921.0	1730.9 ^b	1678.4 ^b	292.8 ^b	222.2 ^b	214.4	95.0 ^b	93.8 ^b	27.5 ^b	14.1 ^b	14.1 ^b
37 Rb	15199.7	2065.1	1863.9	1804.4	326.7 ^b	248.7 ^b	239.1 ^b	113.0 ^b	112 ^b	30.5 ^b	16.3 ^b	15.3 ^b
38 Sr	16104.6	2216.3	2006.8	1939.6	358.7 ^c	280.3 ^c	270.0 ^c	136.0 ^c	134.2 ^c	38.9 ^c	20.3 ^c	20.3 ^c
39 Y	17038.4	2372.5	2155.5	2080.0	392.0 ^b	310.6 ^b	298.8 ^b	157.7 ^c	155.8 ^c	43.8 ^b	24.4 ^b	23.1 ^b
40 Zr	17997.6	2531.6	2306.7	2222.3	430.3 ^c	343.5 ^c	329.8 ^c	181.1 ^c	178.8 ^c	50.6 ^c	28.5 ^c	27.7 ^c
41 Nb	18985.6	2697.7	2464.7	2370.5	466.6 ^c	376.1 ^c	360.6 ^c	205.0 ^c	202.3 ^c	56.4 ^c	32.6 ^c	30.8 ^c
42 Mo	19999.5	2865.5	2625.1	2520.2	506.3 ^c	411.6 ^c	394.0 ^c	231.1 ^c	227.9 ^c	63.2 ^c	37.6 ^c	35.5 ^c
43 Tc	21044.0	3042.5	2793.2	2676.9	544 ^b	445 ^b	425 ^b	257 ^b	253 ^b	68 ^b	39 ^c	39 ^b
44 Ru	22117.2	3224.0	2966.9	2837.9	586.2 ^c	483.5 ^c	461.4 ^c	284.2 ^c	280.0 ^c	75.0 ^c	46.5 ^c	43.2 ^c
45 Rh	23219.9	3411.9	3146.1	3003.8	628.1 ^c	521.3 ^c	496.5 ^c	311.9 ^c	307.2 ^c	81.4 ^b	50.5 ^c	47.3 ^c
46 Pd	24350.3	3604.3	3330.3	3173.3	671.6 ^c	559.9 ^c	532.3 ^c	340.5 ^c	335.2 ^c	87.6 ^b	55.7 ^c	50.9 ^c
47 Ag	25514.0	3805.8	3523.7	3351.1	719.0 ^c	603.8 ^c	573.0 ^c	374.0 ^c	368.0 ^c	97.0 ^c	63.7 ^c	58.3 ^c

[www\(cxro.lbl.gov](http://www.cxro.lbl.gov)



Broadly Tunable Radiation is Needed to Probe the Primary ($n = 1$ & $n = 2$) Resonances of the Elements



Group IA

1	1.0079	1	H
		1s¹	IIA
		Hydrogen	
3	6.941	4	9.012
0.53		1.85	2
4.63	Li	12.3	Be
		22.3	
	1s² 2s¹	18 ² 2s ²	
	Lithium		Beryllium
11	22.990	12	24.31
0.97		1.74	2
2.53	Na	4.30	Mg
		3.20	
	[Ne]3s ¹	[Ne]3s ²	
	Sodium		Magnesium
19	39.098	20	40.08
0.86		1.53	2
1.33	K	2.30	Ca
		3.95	
	[Ar]4s ¹	[Ar]4s ²	
	Potassium		Calcium
37	85.47	38	87.62
1.53		2.58	2
1.08	Rb	1.78	Sr
		4.30	
	[Kr]5s ¹	[Kr]5s ²	
	Rubidium		Strontium
55	132.91	56	137.33
1.90		3.59	
0.86	Cs	1.58	Ba
		4.35	
	[Xe]6s ¹	[Xe]6s ²	
	Cesium		Barium
87	(223)	88	(226)
	1		2
	Fr		Ra
	[Rn]7s ¹	[Rn]7s ²	
	Francium		Radium

Lanthanide series

Actinide series

The diagram shows the periodic table entry for Silicon (Si). The element symbol 'Si' is in the center, with the atomic number 14 above it and the atomic weight 28.09 below it. To the left of the symbol, the electron configuration [Ne]3s²3p² is listed, along with the name 'Silicon'. Arrows point from various labels to specific parts of the element box:

- Atomic number:** Points to the value 14.
- Atomic weight:** Points to the value 28.09.
- Oxidation states:** Points to the superscript '4' and the text '(Bold most stable)'.
- Symbol:** Points to the element symbol 'Si'.
- Density (g/cm³):** Points to the values 2.33, 4.99, and 2.35.
- Concentration (10²²atoms/cm³):** Points to the values 2.33, 4.99, and 2.35.
- Nearest neighbor (Å):** Points to the values 2.33, 4.99, and 2.35.
- Name:** Points to the word 'Silicon' below the electron configuration.
- Electron configuration:** Points to the text '[Ne]3s²3p²'.
- Key:** A legend on the right side defines four categories:
 - Solid (black square)
 - Gas (blue square)
 - Liquid (green square)
 - Synthetically prepared (red square)

References: International Tables for X-ray Crystallography (Reidel, London, 1983) (Ref. 44) and J.R. De Laeter and K.G. Heumann (Ref. 46, 1991).

		VIII			
		2		4.003	
		He		1	
		1s ² Helium			
		IIIIB	IVB	VB	VIB
5	10.81 3 2.47 13.7 1.78 1s ² 2s ¹ 2p ¹ Boron	6 2.27 11.4 1.42 1s ² 2s ¹ 2p ¹ Carbon	7 14.007 ±4.2 1s ² 2s ¹ 2p ³ N	8 -16 -2 1s ² 2s ¹ 2p ⁴ Oxygen	9 19.00 -1 1s ² 2s ¹ 2p ⁵ Fluorine
13	26.98 3 2.70 6.02 2.86 [Ne]3s ² 3p ¹ Aluminum	14 28.09 4 2.33 4.99 2.35 [Ne]3s ² 3p ² Silicon	15 30.974 ±3.5.4 1.82 3.54 [Ne]3s ² 3p ³ Phosphorus	16 32.066 ±2.4.6 2.09 3.92 [Ne]3s ² 3p ⁴ Sulfur	17 35.453 ±1.3.5.7 [Ne]3s ² 3p ⁵ Chlorine
31	69.72 3 5.91 5.10 2.44 [Ar]3d ¹⁰ 4s ² 4p ¹ Gallium	32 72.61 4 5.32 4.42 2.45 [Ar]3d ¹⁰ 4s ² 4p ² Germanium	33 74.92 ±3.5 5.78 4.64 2.51 [Ar]3d ¹⁰ 4s ² 4p ³ Arsenic	34 78.96 -2.4.6 4.81 3.67 2.32 [Ar]3d ¹⁰ 4s ² 4p ⁴ Selenium	35 79.904 ±1.5 3.12 2.35 [Ar]3d ¹⁰ 4s ² 4p ⁵ Bromine
41	114.82 3 7.29 3.62 3.25 [Kr]4d ¹⁰ 5s ² 5p ¹ Indium	50 118.71 4.2 7.29 3.70 3.02 [Kr]4d ¹⁰ 5s ² 5p ² Tin	51 121.76 ±3.5 6.69 3.31 2.90 [Kr]4d ¹⁰ 5s ² 5p ³ Antimony	52 127.60 -2.4.6 6.25 5.95 2.86 [Kr]4d ¹⁰ 5s ² 5p ⁴ Tellurium	53 126.90 ±1.5.7 4.95 2.35 [Kr]4d ¹⁰ 5s ² 5p ⁵ Iodine
59	204.38 3.1 11.9 3.50 3.25 [Xe]4f ¹⁴ 5d ¹ 6s ² 6p ¹ Thallium	82 207.2 4.2 11.3 3.30 3.50 [Xe]4f ¹⁴ 5d ¹ 6s ² 6p ² Lead	83 208.98 3.5 9.80 3.69 3.31 [Xe]4f ¹⁴ 5d ¹ 6s ² 6p ³ Bismuth	84 (209) 4.2 9.27 9.67 3.35 [Xe]4f ¹⁴ 5d ¹ 6s ² 6p ⁴ Polonium	85 (210) ±1.3.5.7 4.95 2.35 [Xe]4f ¹⁴ 5d ¹ 6s ² 6p ⁵ Astatine
77)					
50	164.93 3 8.80 3.21 3.49 [Xe]4f ¹¹ 6s ² Holmium	68 167.26 3 9.04 3.26 3.47 [Xe]4f ¹¹ 6s ² Erbium	69 168.93 3.2 9.33 3.32 3.45 [Xe]4f ¹¹ 6s ² Thulium	70 173.04 3.2 6.97 2.42 3.88 [Xe]4f ¹¹ 6s ² Ytterbium	71 174.97 3 9.84 3.39 3.43 [Xe]4f ¹¹ 6s ² Lutetium
51	252 3 99 [Rn]5f ¹¹ 7s ² Einsteinium	100 (257) [Rn]5f ¹² 7s ² Fermium	101 (258) [Rn]5f ¹¹ 7s ² Mendelevium	102 (259) [Rn]5f ¹¹ 7s ² Nobelium	103 (262) [Rn]5f ¹¹ 6d ¹ 7s ² Lawrencium



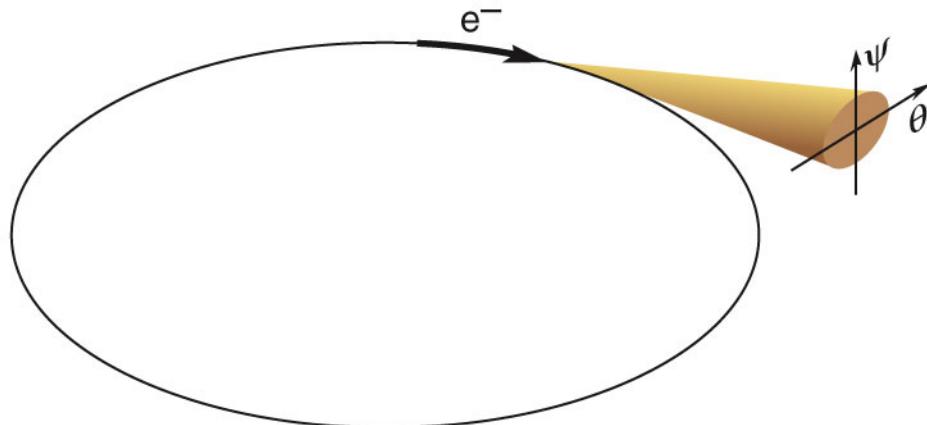
Typical Applications of Synchrotron Radiation



- Surface science
- Magnetic materials
- Materials chemistry
- Environmental sciences
- Protein crystallography
- Biomicroscopy
- Chemical dynamics

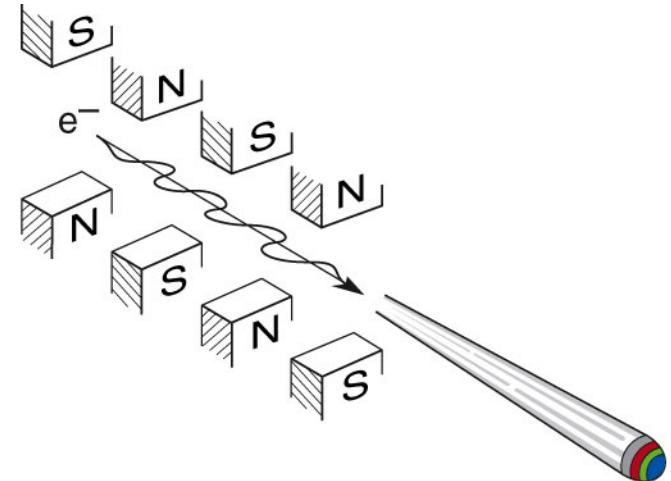


Bright and Powerful X-Rays from Relativistic Electrons



Synchrotron radiation

- 10^{10} brighter than the most powerful (compact) laboratory source
- An x-ray “light bulb” in that it radiates all “colors” (wavelengths, photons energies)



Undulator radiation

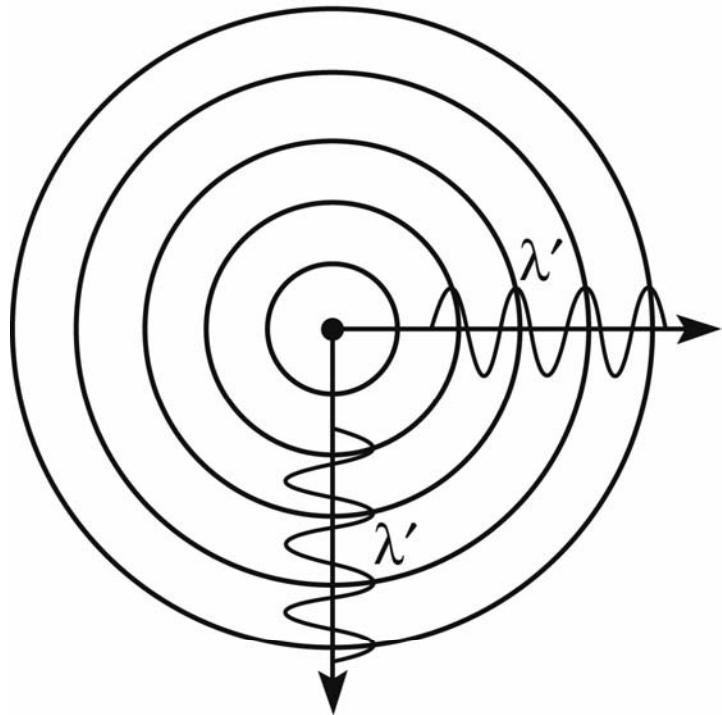
- Lasers exist for the IR, visible, UV, VUV, and EUV
- Undulator radiation is quasi-monochromatic and highly directional, approximating many of the desired properties of an x-ray laser



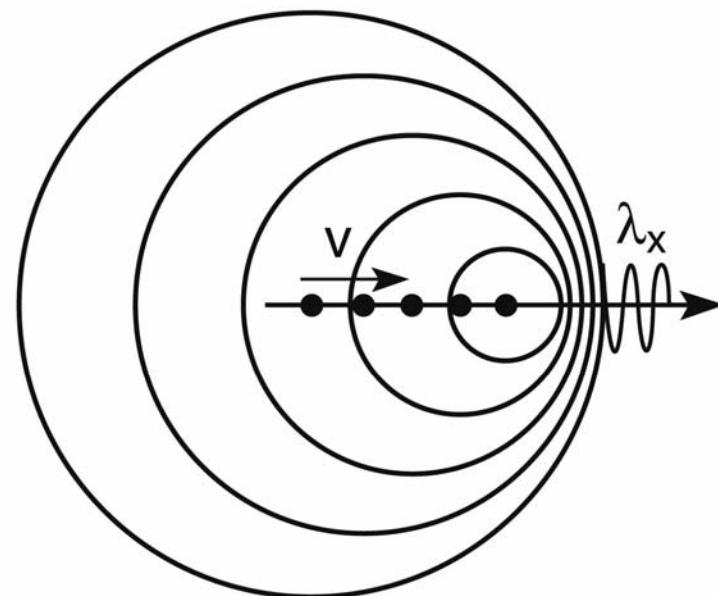
Synchrotron Radiation from Relativistic Electrons



$v \ll c$



$v \lesssim c$



Note: Angle-dependent doppler shift

$$\lambda = \lambda' (1 - \frac{v}{c} \cos\theta)$$

$$\lambda = \lambda' \gamma (1 - \frac{v}{c} \cos\theta)$$

$$\gamma = \frac{1}{\sqrt{1 - \frac{v^2}{c^2}}}$$



Some Useful Formulas for Synchrotron Radiation



$$\gamma = \frac{1}{\sqrt{1 - \frac{v^2}{c^2}}} = \frac{1}{\sqrt{1 - \beta^2}} ; \quad \beta \equiv \frac{v}{c}$$

$$\gamma = \frac{E_e}{mc^2} = 1957 E_e(\text{GeV})$$

$$E_e = \gamma mc^2, \quad \mathbf{p} = \gamma m\mathbf{v}$$

$$\hbar\omega \cdot \lambda = 1239.842 \text{ eV} \cdot \text{nm}$$

$$1 \text{ watt} \Rightarrow 5.034 \times 10^{15} \lambda[\text{nm}] \frac{\text{photons}}{\text{s}}$$

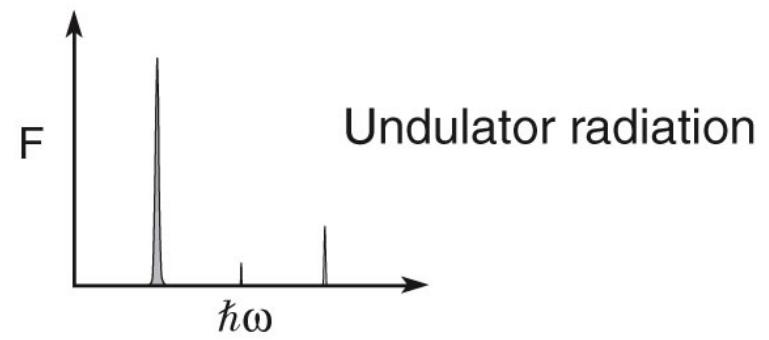
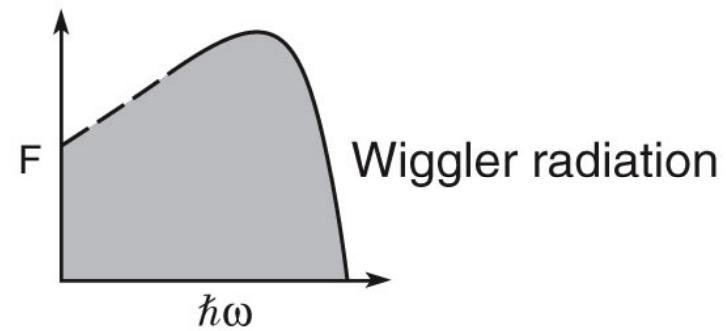
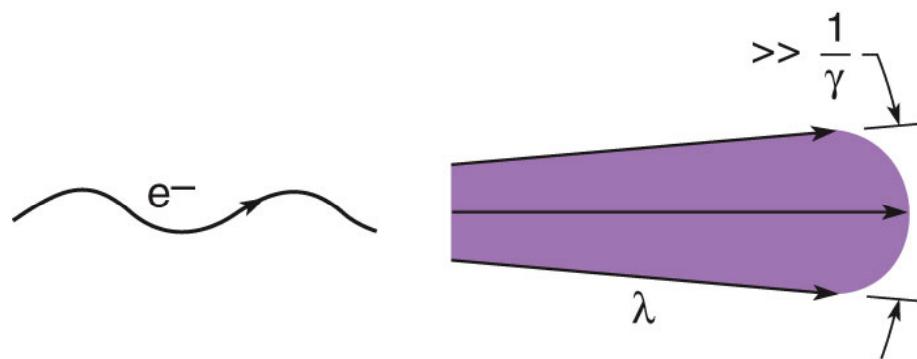
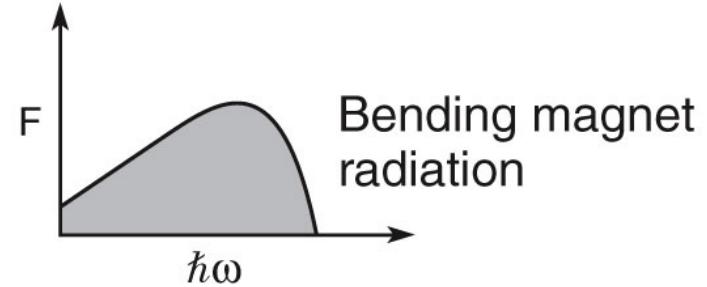
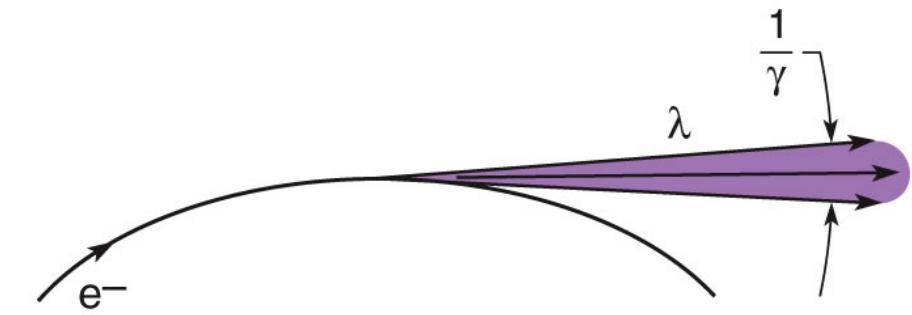
$$\text{Bending Magnet: } E_c = \frac{3e\hbar B\gamma^2}{2m}, \quad E_c(\text{keV}) = 0.6650 E_e^2(\text{GeV}) B(\text{T})$$

$$\text{Undulator: } \lambda = \frac{\lambda_u}{2\gamma^2} \left(1 + \frac{K^2}{2} + \gamma^2 \theta^2 \right); \quad E(\text{keV}) = \frac{0.9496 E_e^2(\text{GeV})}{\lambda_u(\text{cm}) \left(1 + \frac{K^2}{2} + \gamma^2 \theta^2 \right)}$$

$$\text{where } K \equiv \frac{eB_0\lambda_u}{2\pi mc} = 0.9337 B_0(\text{T}) \lambda_u(\text{cm})$$

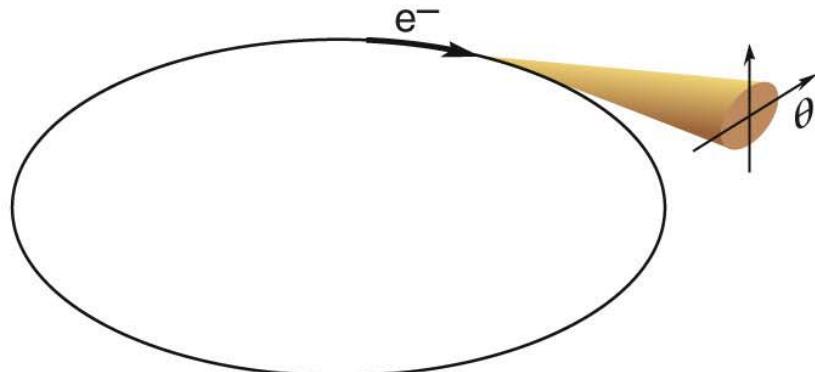


Three Forms of Synchrotron Radiation





Bending Magnet Radiation Covers a Broad Region of the Spectrum, Including the Primary Absorption Edges of Most Elements



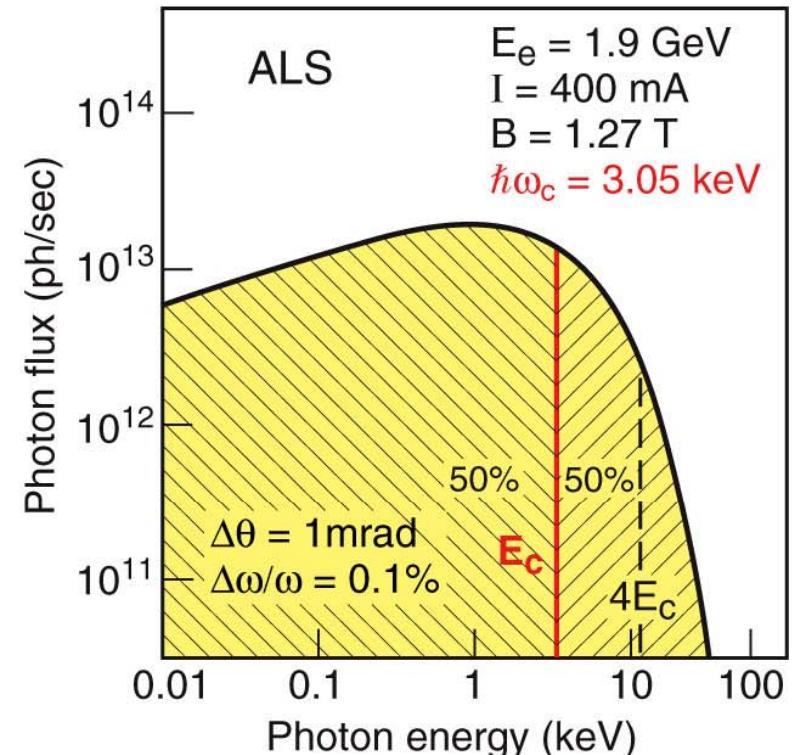
$$E_c = \hbar\omega_c = \frac{3e\hbar B\gamma^2}{2m} \quad (5.7a)$$

$$E_c(\text{keV}) = 0.6650 E_e^2 (\text{GeV}) B (\text{T}) \quad (5.7b)$$

$$\frac{d^2 F_B}{d\theta d\omega/\omega} = 2.46 \times 10^{13} E_e (\text{GeV}) I (\text{A}) G_1(E/E_c) \frac{\text{photons/s}}{\text{mrad} \cdot (0.1\% \text{BW})} \quad (5.8)$$

- Advantages:
- covers broad spectral range
 - least expensive
 - most accessible

- Disadvantages:
- limited coverage of hard x-rays
 - not as bright as undulator

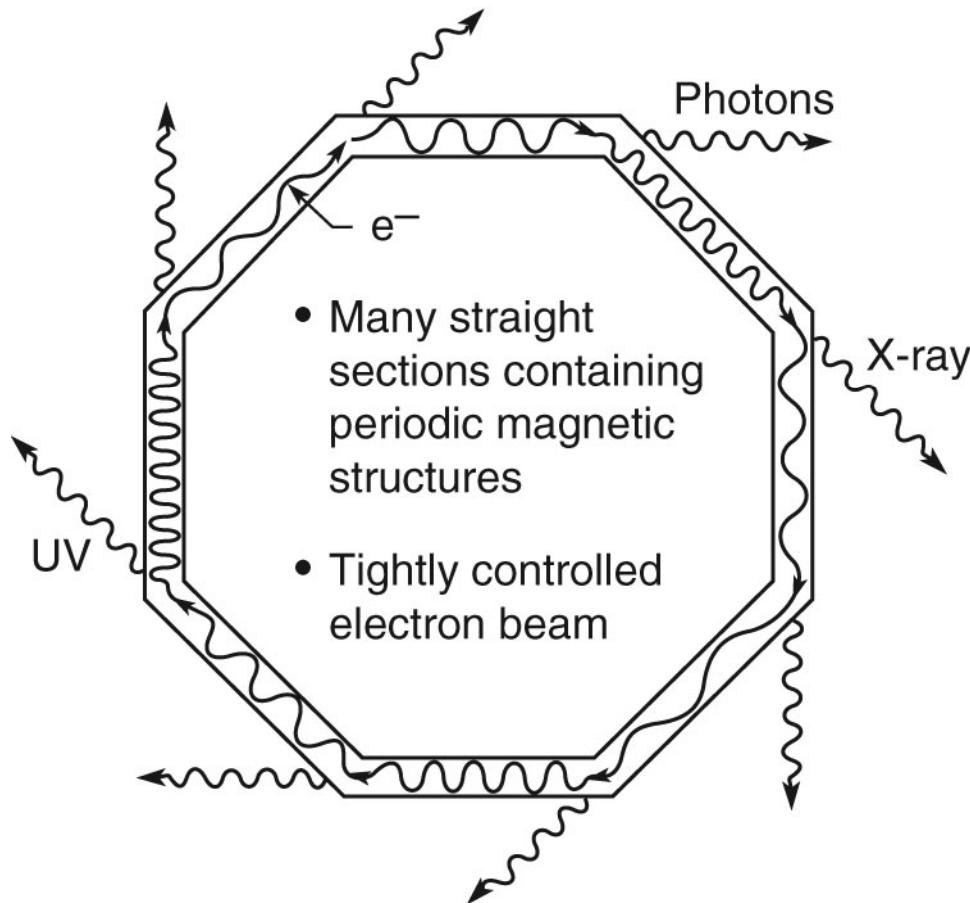




Third Generation Facilities, Like Elettra, Have Many Straight Sections and a Small Electron Beam



Modern Synchrotron Radiation Facility



- Many straight sections for undulators and wigglers
- Brighter radiation for spatially resolved studies (smaller beam more suitable for microscopies)
- Interesting coherence properties at very short wavelengths

3rdGenFacilitiesrev3.04.ai

Third Generation Synchrotron Facilities

ESRF	6 GeV	France
ALS	1.9 GeV	USA
APS	7 GeV	USA
BESSY II	1.7 GeV	Germany
ELETTRA	2.0 GeV	Italy
SPring-8	8 GeV	Japan
MAX II	1.5 GeV	Sweden
SLS	2.4 GeV	Switzerland
PLS	2 GeV	Korea
SRRCC	1.4 GeV	Taiwan
SSRL	3 GeV	USA
CLS	2.9 GeV	Canada

Third Generation Synchrotron Facilities Under Construction

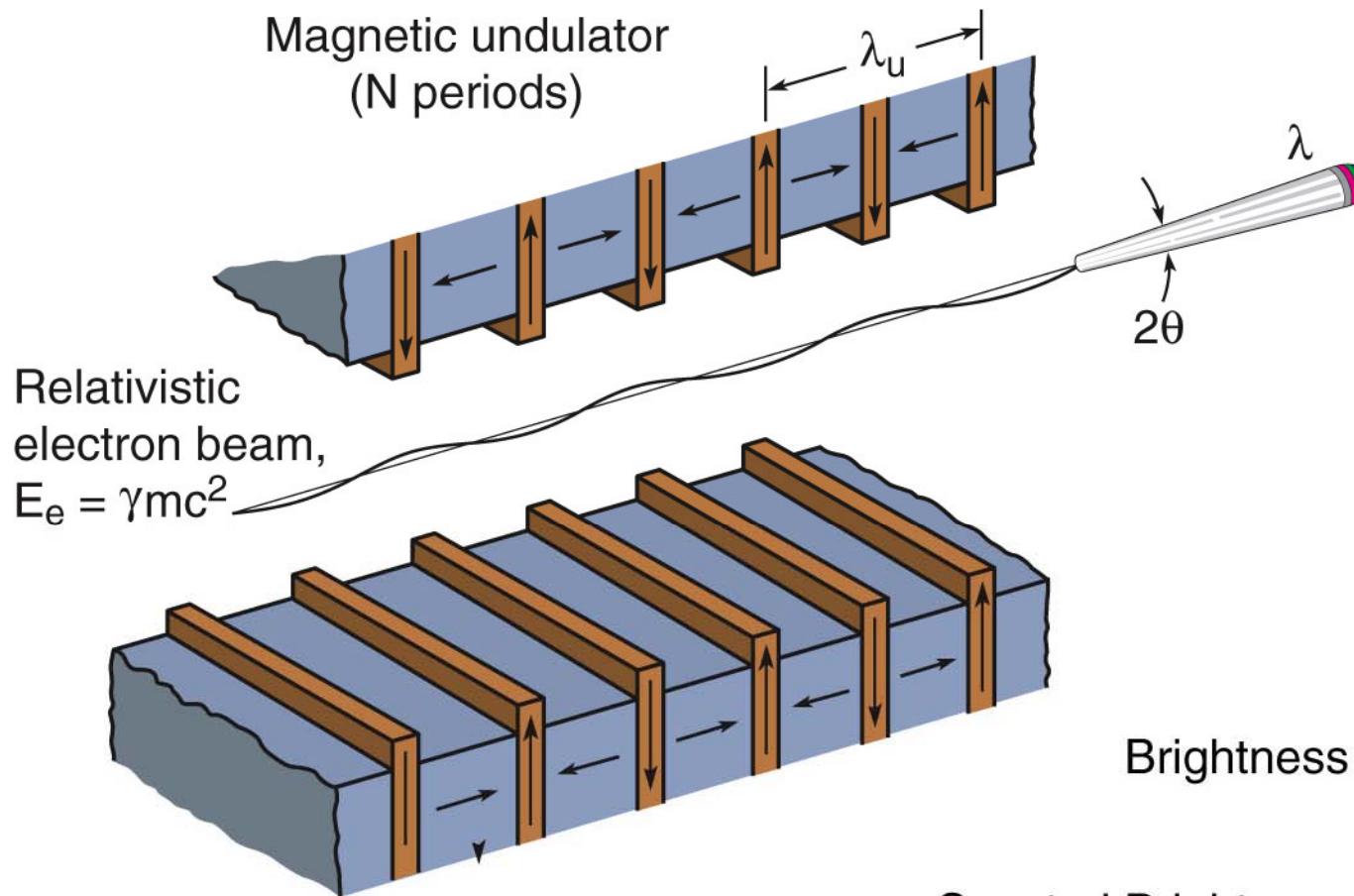
Soleil	2.5 GeV	France
Diamond	3 GeV	UK
Australian Light Source	3 GeV	Australia

Others are in the design stage or planning an upgrade to third generation.

Courtesy of Herman Winick, SSRL, Stanford
www-srsl.slac.stanford.edu/SR_SOURCES.HTML



Undulator Radiation from a Small Electron Beam Radiating into a Narrow Forward Cone is Very Bright



$$\lambda \approx \frac{\lambda_u}{2\gamma^2}$$

$$\theta_{cen} \approx \frac{1}{\gamma \sqrt{N}}$$

$$\left[\frac{\Delta\lambda}{\lambda} \right]_{cen} = \frac{1}{N}$$

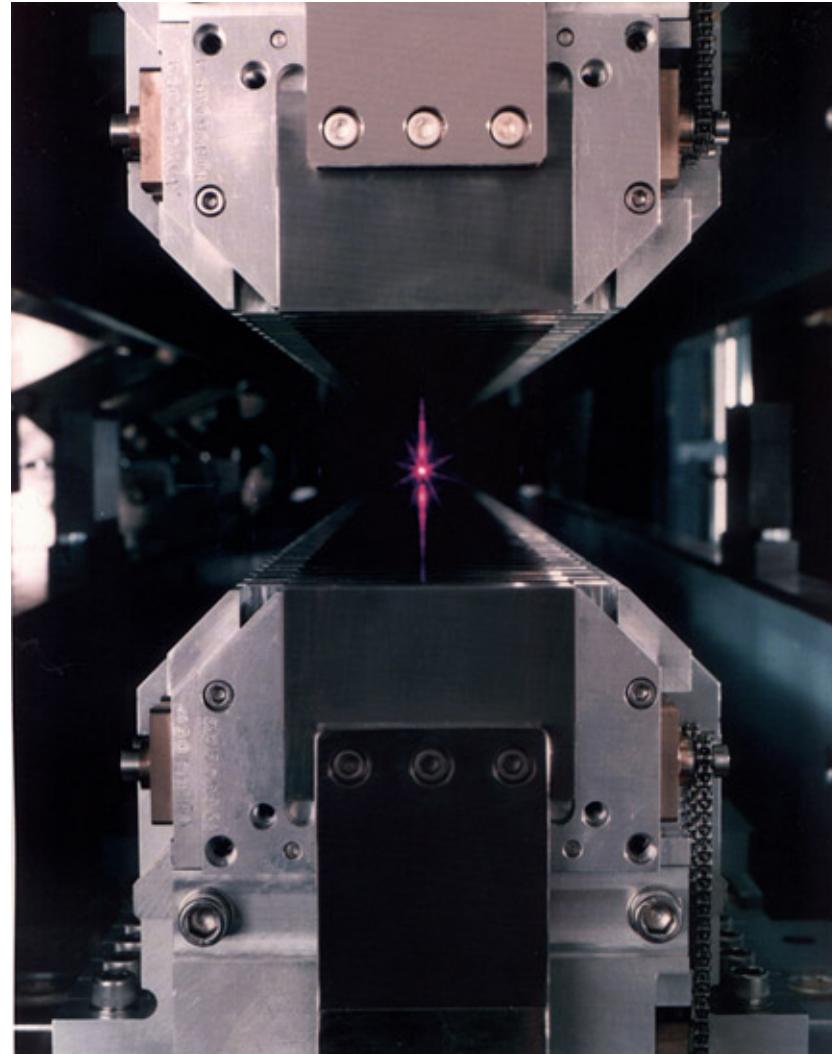
$$\text{Brightness} = \frac{\text{photon flux}}{(\Delta A) (\Delta \Omega)}$$

$$\text{Spectral Brightness} = \frac{\text{photon flux}}{(\Delta A) (\Delta \Omega) (\Delta \lambda / \lambda)}$$

Ch05_F08VG_1.04.ai



An Undulator Up Close



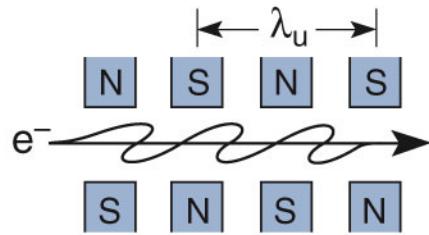
ALS U5 undulator, beamline 7.0,
 $N = 89, \lambda_u = 50 \text{ mm}$



Undulator Radiation



Laboratory Frame of Reference

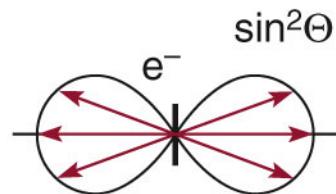


$$E = \gamma mc^2$$

$$\gamma = \frac{1}{\sqrt{1 - \frac{v^2}{c^2}}}$$

N = # periods

Frame of Moving e^-



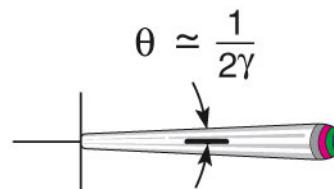
e^- radiates at the Lorentz contracted wavelength:

$$\lambda' = \frac{\lambda_u}{\gamma}$$

Bandwidth:

$$\frac{\lambda'}{\Delta\lambda} \simeq N$$

Frame of Observer



Doppler shortened wavelength on axis:

$$\lambda = \lambda' \gamma (1 - \beta \cos \theta)$$

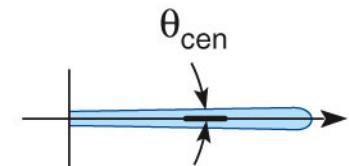
$$\lambda = \frac{\lambda_u}{2\gamma^2} (1 + \gamma^2 \theta^2)$$

Accounting for transverse motion due to the periodic magnetic field:

$$\lambda = \frac{\lambda_u}{2\gamma^2} \left(1 + \frac{K^2}{2} + \gamma^2 \theta^2 \right)$$

where $K = eB_0\lambda_u / 2\pi mc$

Following Monochromator



$$\text{For } \frac{\Delta\lambda}{\lambda} \simeq \frac{1}{N}$$

$$\theta_{\text{cen}} \simeq \frac{1}{\gamma \sqrt{N}}$$

typically

$$\theta_{\text{cen}} \simeq 40 \mu\text{rad}$$



Power in the Central Cone



$$\lambda_x = \frac{\lambda_u}{2\gamma^2} \left(1 + \frac{K^2}{2} + \gamma^2 \theta^2\right)$$

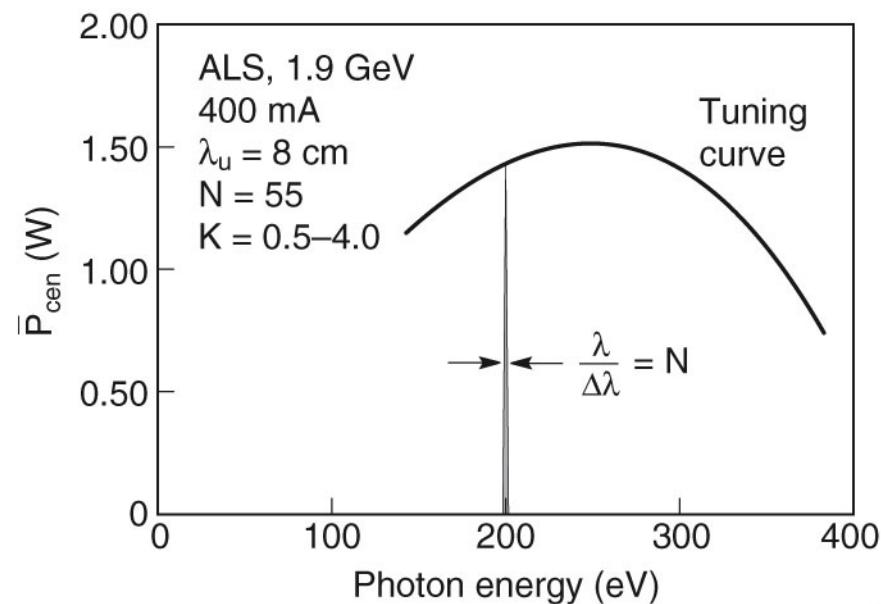
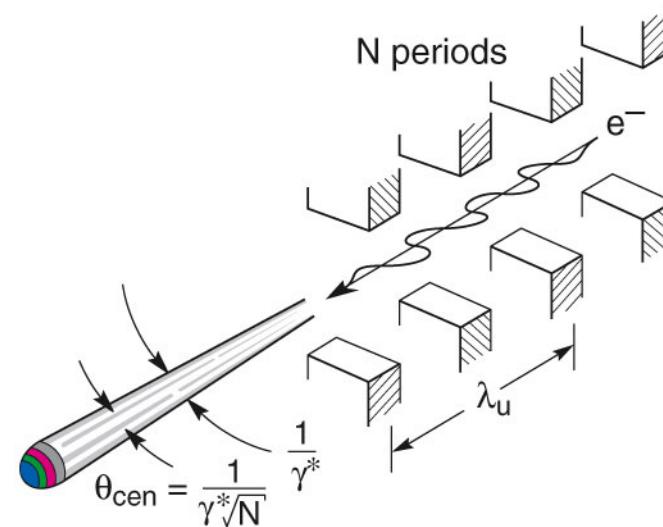
$$\bar{P}_{cen} = \frac{\pi e \gamma^2 I}{\epsilon_0 \lambda_u} \frac{K^2}{(1 + \frac{K^2}{2})^2} f(K)$$

$$\theta_{cen} = \frac{1}{\gamma^* \sqrt{N}}$$

$$\left(\frac{\Delta\lambda}{\lambda}\right)_{cen} = \frac{1}{N}$$

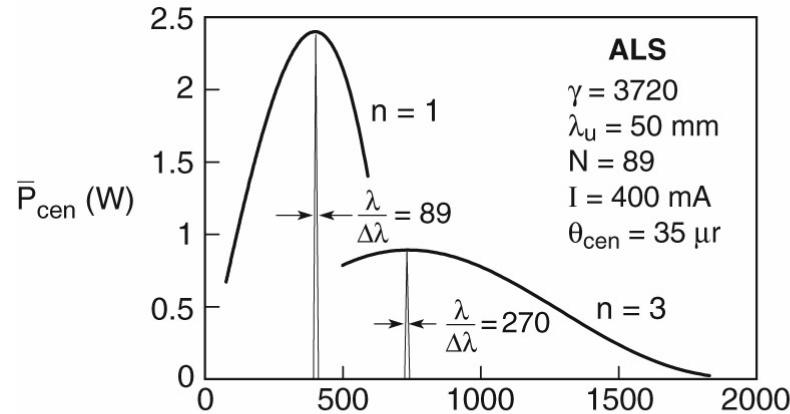
$$K = \frac{eB_0\lambda_u}{2\pi m_0 c}$$

$$\gamma^* = \gamma / \sqrt{1 + \frac{K^2}{2}}$$





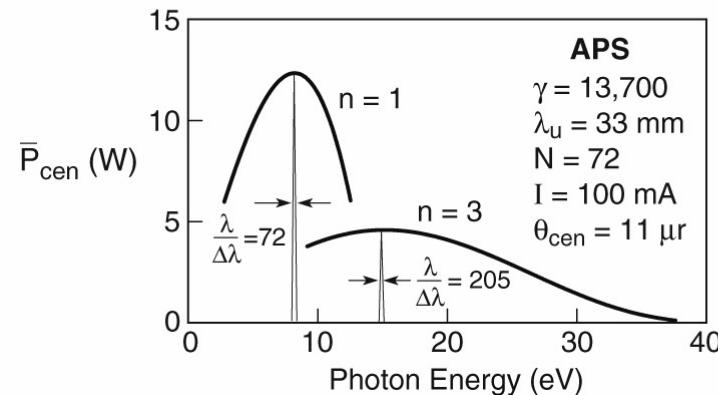
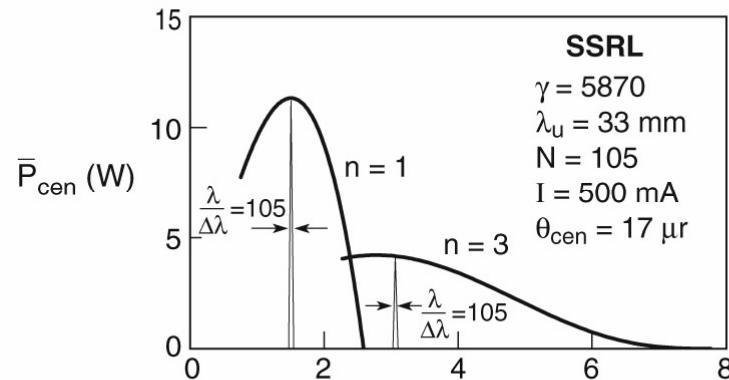
Power in the Central Radiation Cone for Three Soft X-Ray Undulators



$$\theta_{cen} = \frac{1}{\gamma^* \sqrt{N}}$$

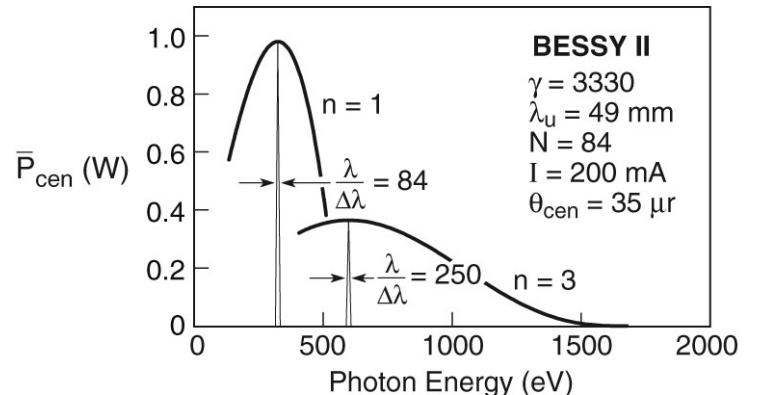
$$\left[\frac{\Delta\lambda}{\lambda} \right]_1 = \frac{1}{N}$$

$$\left[\frac{\Delta\lambda}{\lambda} \right]_3 = \frac{1}{3N}$$





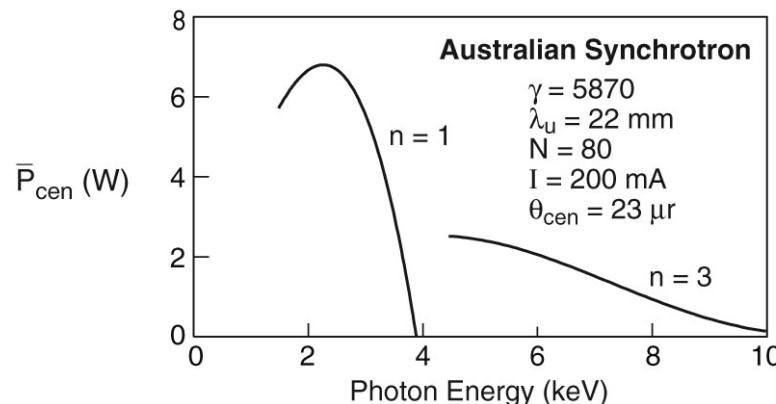
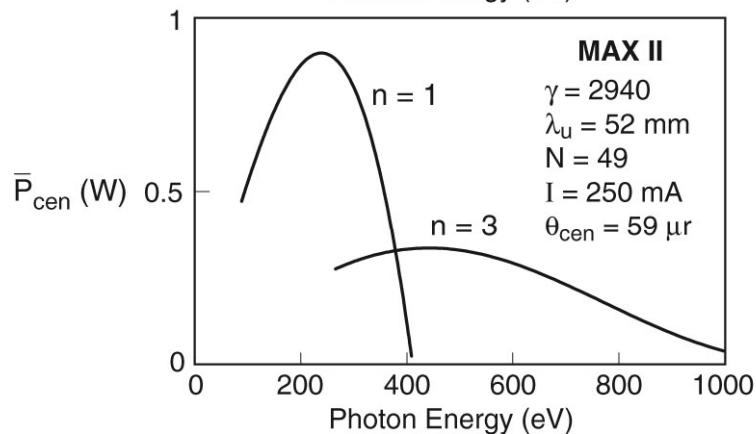
Power in the Central Radiation Cone for Three Soft X-Ray Undulators



$$\theta_{cen} = \frac{1}{\gamma^* \sqrt{N}}$$

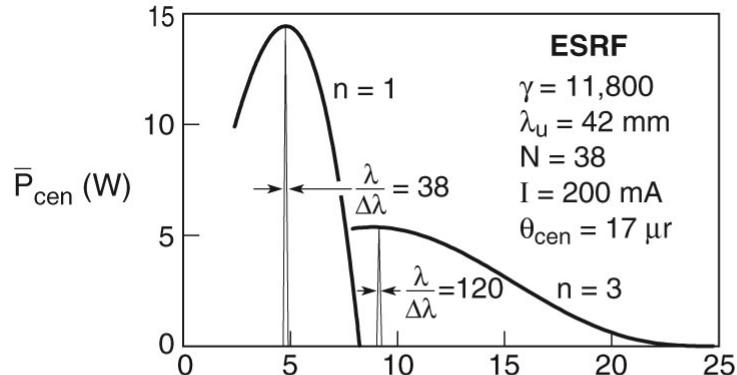
$$\left[\frac{\Delta\lambda}{\lambda} \right]_1 = \frac{1}{N}$$

$$\left[\frac{\Delta\lambda}{\lambda} \right]_3 = \frac{1}{3N}$$





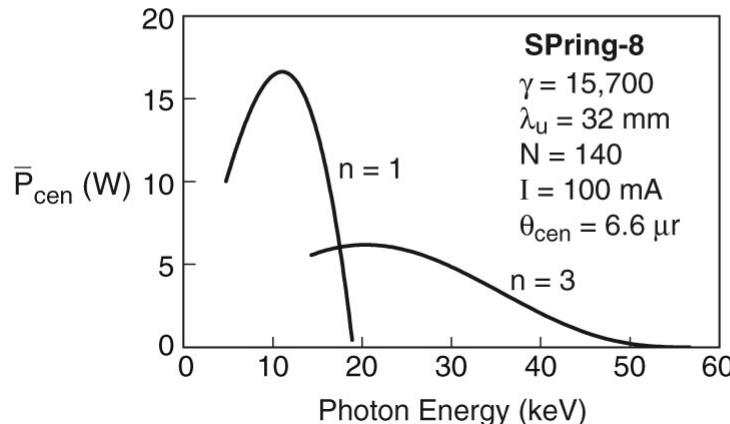
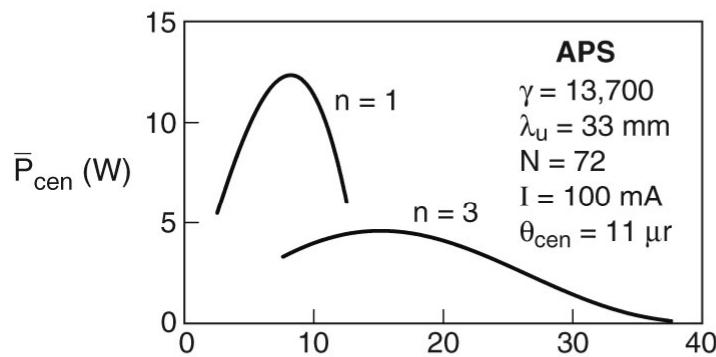
Power in the Central Radiation Cone for Three Soft X-Ray Undulators



$$\theta_{cen} = \frac{1}{\gamma^* \sqrt{N}}$$

$$\left[\frac{\Delta\lambda}{\lambda} \right]_1 = \frac{1}{N}$$

$$\left[\frac{\Delta\lambda}{\lambda} \right]_3 = \frac{1}{3N}$$





Brightness and Spectral Brightness



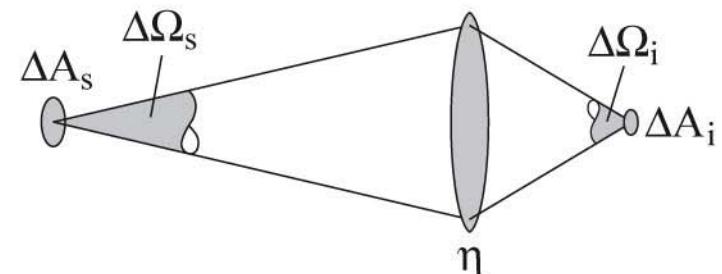
Brightness is defined as radiated power per unit area and per unit solid angle at the source:

$$B = \frac{\Delta P}{\Delta A \cdot \Delta \Omega} \quad (5.57)$$

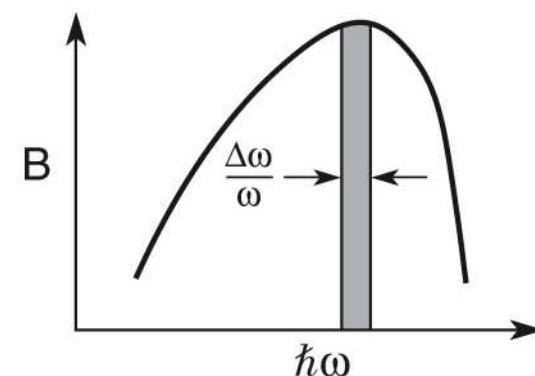
Brightness is a conserved quantity in perfect optical systems, and thus is useful in designing beamlines and synchrotron radiation experiments which involve focusing to small areas.

Spectral brightness is that portion of the brightness lying within a relative spectral bandwidth $\Delta\omega/\omega$:

$$B_{\Delta\omega/\omega} = \frac{\Delta P}{\Delta A \cdot \Delta \Omega \cdot \Delta\omega/\omega} \quad (5.58)$$



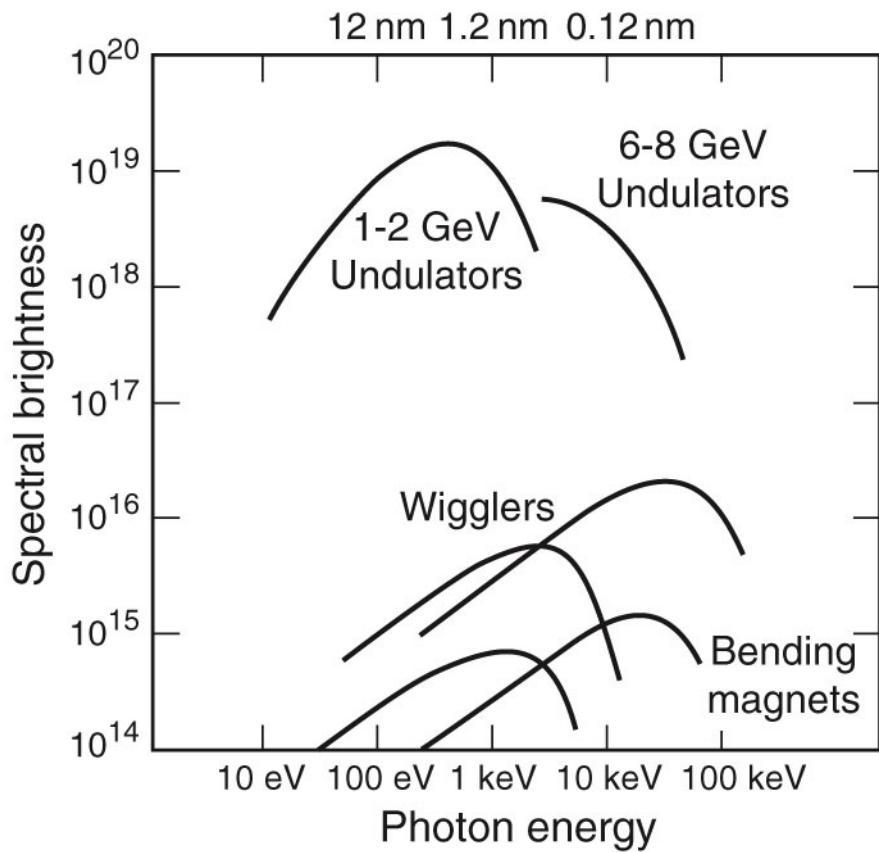
Perfect optical system:
 $\Delta A_s \cdot \Delta \Omega_s = \Delta A_i \cdot \Delta \Omega_i ; \eta = 100\%$



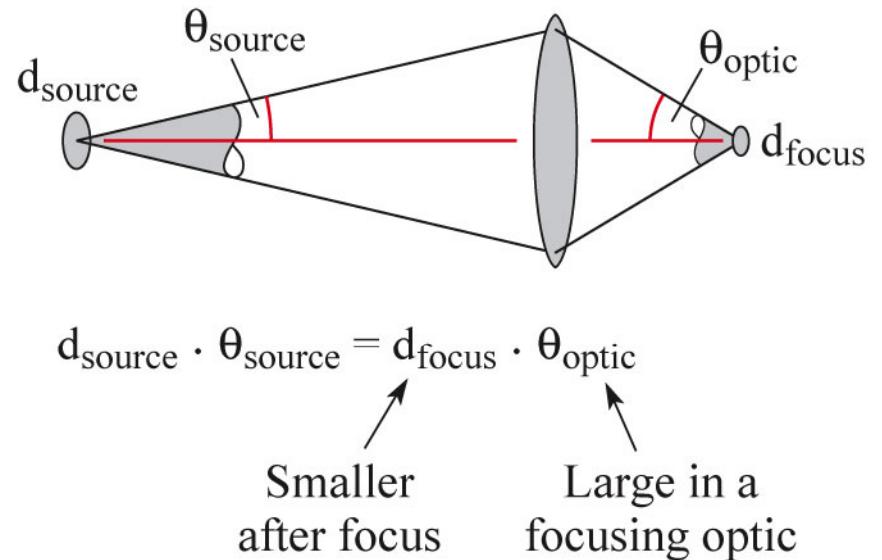
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Spectral Brightness is Useful for Experiments that Involve Spatially Resolved Studies



- Brightness is conserved (in lossless optical systems)

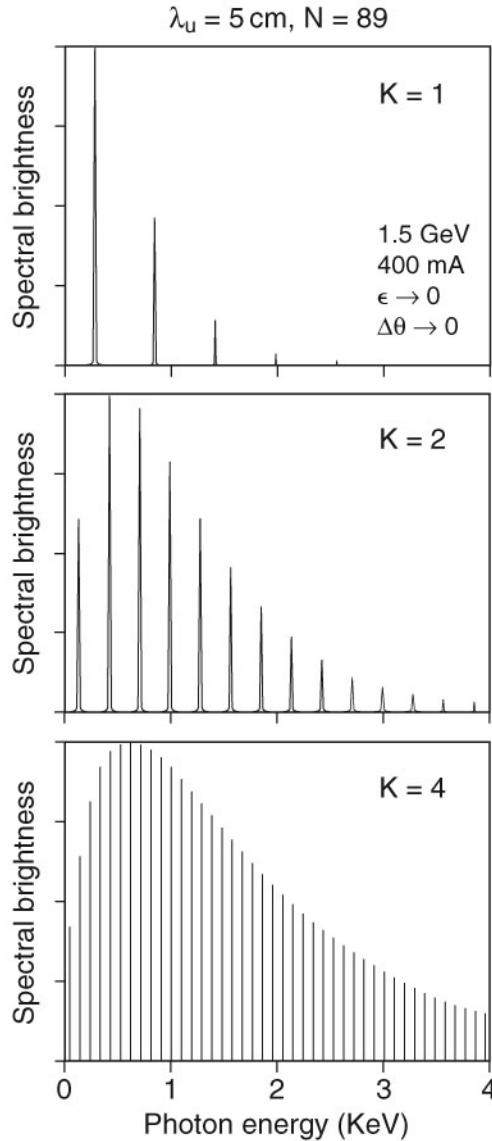


- Starting with many photons in a small source area and solid angle, permits high photon flux in an even smaller area

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The Transition from Undulator Radiation ($K \leq 1$) to Wiggler Radiation ($K \gg 1$)



Undulator radiation ($K \lesssim 1$)

- Narrow spectral lines
- High spectral brightness
- Partial coherence

$$\lambda = \frac{\lambda_u}{2\gamma^2} \left(1 + \frac{K^2}{2} + \gamma^2 \theta^2 \right)$$

$$K = \frac{eB_0\lambda_u}{2\pi mc}$$

Wiggler radiation ($K \gg 1$)

- Higher photon energies
- Spectral continuum
- Higher photon flux ($2N$)

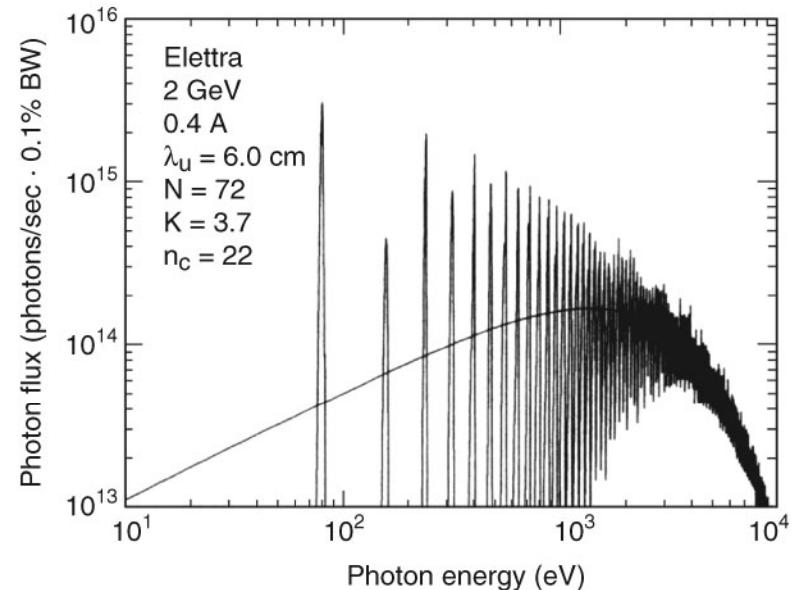
$$\hbar\omega_c = \frac{3}{2} \frac{\hbar\gamma^2 e B_0}{m}$$

$$n_c = \frac{3K}{4} \left(1 + \frac{K^2}{2} \right)$$

(Courtesy of K.-J. Kim)

D. Attwood

24



(Courtesy of R.P. Walker and B. Diviacco)

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IntroSynchRad_Apps2005.ppt

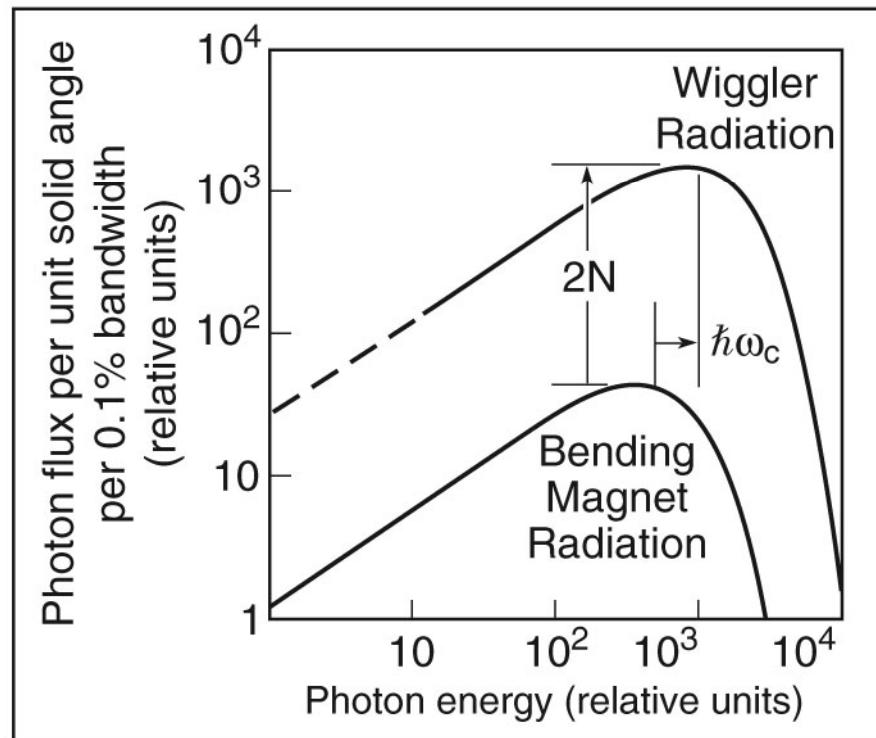


Wiggler Radiation



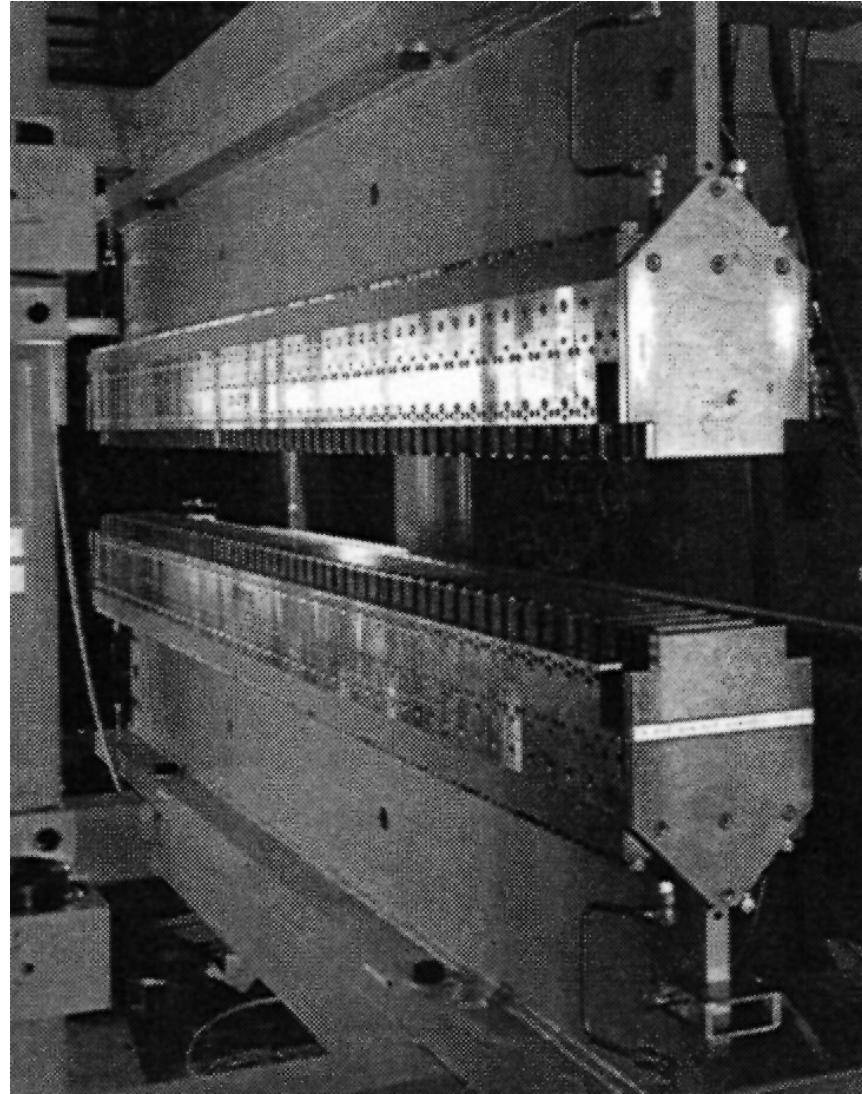
At very high $K \gg 1$, the radiated energy appears in very high harmonics, and at rather large horizontal angles $\theta \approx \pm K/\gamma$ (eq. 5.21). Because the emission angles are large, one tends to use larger collection angles, which tends to spectrally merge nearby harmonics. The result is a continuum at very high photon energies, similar to that of bending magnet radiation, but increased by $2N$ (the number of magnet pole pieces).

$$E_c = \hbar\omega_c = \frac{3e\hbar B\gamma^2}{2m} ; \quad n_c = \frac{3K}{4} \left(1 + \frac{K^2}{2}\right) \quad (5.7a \& 82)$$





Stanford Permanent Magnet Wiggler



LBNL/EXXON/SSRL (1982), SSRL Beamline VI
55 pole ($N = 27.5$), $\lambda_w = 7$ cm



Typical Parameters for Synchrotron Radiation



Facility	ALS	SSRL	APS
Electron energy	1.90 GeV	_____ GeV	7.00 GeV
γ	3720	_____	13,700
Current (mA)	400	_____	100
Circumference (m)	197	_____	1100
RF frequency (MHz)	500	_____	352
Pulse duration (FWHM) (ps)	35–100	_____	170
<i>Bending Magnet Radiation:</i>			
Bending magnet field (T)	1.27	_____	0.599
Critical photon energy (keV)	3.05	_____	19.5
Critical photon wavelength	0.407 nm	_____ nm	0.0636 nm
Bending magnet sources	24	_____	35
<i>Undulator Radiation:</i>			
Number of straight sections	12	_____	40
Undulator period (typical) (cm)	5.00	_____	3.30
Number of periods	89	_____	72
Photon energy ($K = 1, n = 1$)	457 eV	_____ eV	9.40 keV
Photon wavelength ($K = 1, n = 1$)	2.71 nm	_____ nm	0.132 nm
Tuning range ($n = 1$)	2.0–5.4 nm	_____ nm	0.10–0.35 nm
Tuning range ($n = 3$)	0.68–1.8 nm	_____ nm	0.033–0.12 nm
Central cone half-angle ($K = 1$)	35 μ rad	_____ μ rad	11 μ rad
Power in central cone ($K = 1, n = 1$) (W)	2.3	_____	12
Flux in central cone (photons/s)	3.1×10^{16}	____ \times ____	7.9×10^{15}
σ_x, σ_y (μ m)	260, 16	_____	320, 50
σ'_x, σ'_y (μ rad)	23, 3.9	_____	23, 7
Brightness ($K = 1, n = 1$) ^a [(photons/s)/mm ² · mrad ² · (0.1%BW)]	2.3×10^{19}	____ \times ____	4.8×10^{18}
Total power ($K = 1$, all n , all θ) (W)	187	_____	780
Other undulator periods (cm)	3.65, 8.00, 10.0	_____	2.70, 5.50, 12.8
<i>Wiggler Radiation:</i>			
Wiggler period (typical) (cm)	16.0	_____	8.5
Number of periods	19	_____	28
Magnetic field (maximum) (T)	2.1	_____	1.0
K (maximum)	32	_____	7.9
Critical photon energy (keV)	5.1	_____	33
Critical photon wavelength	0.24 nm	_____ nm	0.038 nm (0.38 \AA)
Total power (max. K) (kW)	13	_____	7.4

^aUsing Eq. (5.65). See comments following Eq. (5.64) for the case where $\sigma'_{x,y} \simeq \theta_{cen}$.



Typical Parameters for Synchrotron Radiation

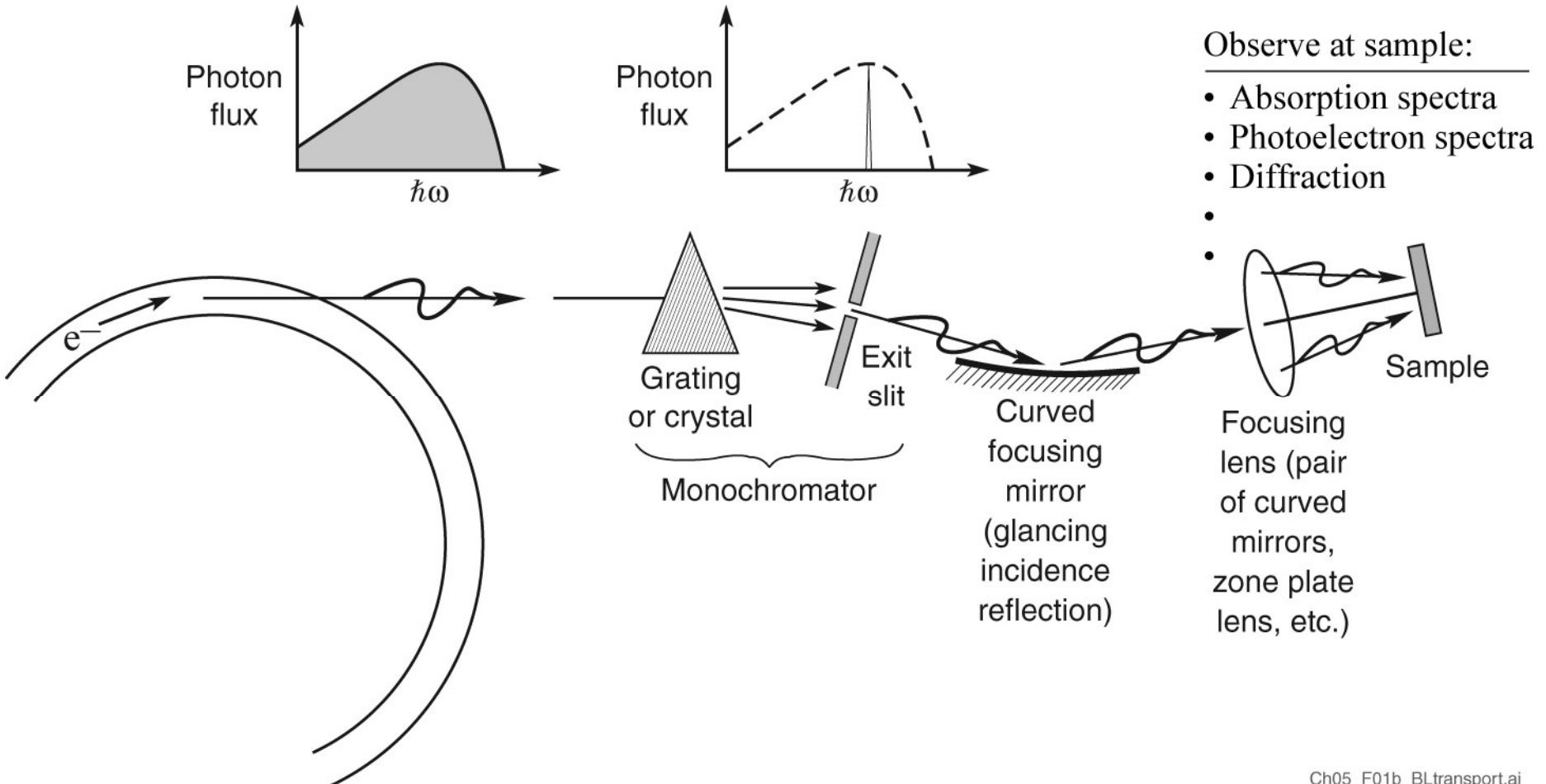


Facility	MAX II	BESSY II	ESRF	SP-8
Electron energy	1.50 GeV	1.70 GeV	6.04 GeV	8.00 GeV
γ	2940	3330	11,800	15,700
Current (mA)	250	200	200	100
Circumference (m)	90	240	884	1440
RF frequency (MHz)	500	500	352	509
Pulse duration (FWHM) (ps)	200	20–50	70	120
<i>Bending Magnet Radiation:</i>				
Bending magnet field (T)	1.48	1.30	0.806	0.679
Critical photon energy (keV)	2.21	2.50	19.6	28.9
Critical photon wavelength	0.560 nm	0.50 nm	0.634 Å	0.429 Å
Bending magnet sources	20	32	32	23
<i>Undulator Radiation:</i>				
Number of straight sections	10	16	32	48
Undulator period (typical) (cm)	5.20	4.90	4.20	3.20
Number of periods	49	84	38	140
Photon energy ($K = 1, n = 1$)	274 eV	373 eV	5.50 keV	12.7 kev
Photon wavelength ($K = 1, n = 1$)	4.53 nm	3.32 nm	0.225 nm	0.979 Å
Tuning range ($n = 1$)	130–410 eV	140–500 eV	2.6–7.3 keV	4.7–19 keV
Tuning range ($n = 3$)	400–1200 eV	410–1100 eV	7.7–22 keV	16–51 keV
Central cone half-angle ($K = 1$)	59 μ rad	33 μ rad	17 μ rad	6.6 μ rad
Power in central cone ($K = 1, n = 1$) (W)	0.88	0.95	14	16
Flux in central cone (photons/s)	2.0×10^{16}	1.6×10^{16}	1.6×10^{16}	7.9×10^{15}
σ_x, σ_y (μ m)	300, 45	314, 24	395, 9.9	380, 6.8
σ'_x, σ'_y (μ rad)	26, 20	18, 12	11, 3.9	16, 1.8
Brightness ($K = 1, n = 1$) ^a [(photons/s)/mm ² · mrad ² · (0.1%BW)]	7.8×10^{17}	4.6×10^{18}	5.1×10^{18}	1.8×10^{20}
Total power ($K = 1$, all n , all θ) (W)	17	32	480	2,000
Other undulator periods (cm)	5.88, 6.60	4.10, 5.60 12.5	2.30, 3.20 5.20, 8.50	2.4, 10.0, 3.7 12.0
<i>Wiggler Radiation:</i>				
Wiggler period (typical) (cm)	17.4	12.5	8.0	12.0
Number of periods	13	32	20	37
Magnetic field (maximum) (T)	1.80	1.15	0.81	1.0
K (maximum)	29.3	12.8	6.0	11
Critical photon energy (keV)	2.69	2.11	20	43
Critical photon wavelength	0.46 nm	0.59 nm	0.62 Å	0.29 Å
Total power (max. K) (kW)	5.9	1.8	4.8	18

^aUsing Eq. (5.65). See comments following Eq. (5.64) for the case where $\sigma'_{x,y} \simeq \theta_{cen}$.



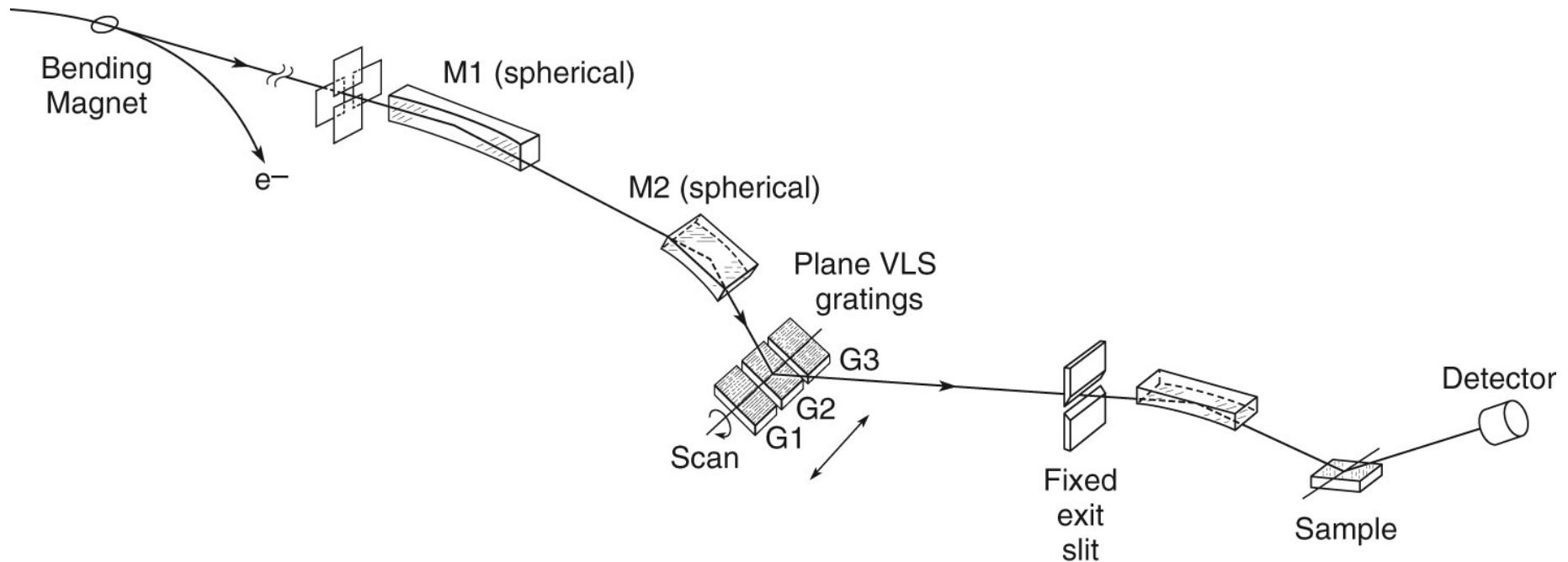
Beamlines are Used to Transport Photons to the Sample, and Take a Desired Spectral Slice



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A Typical Beamline: Monochromator Plus Focusing Optics to Deliver Radiation to the Sample

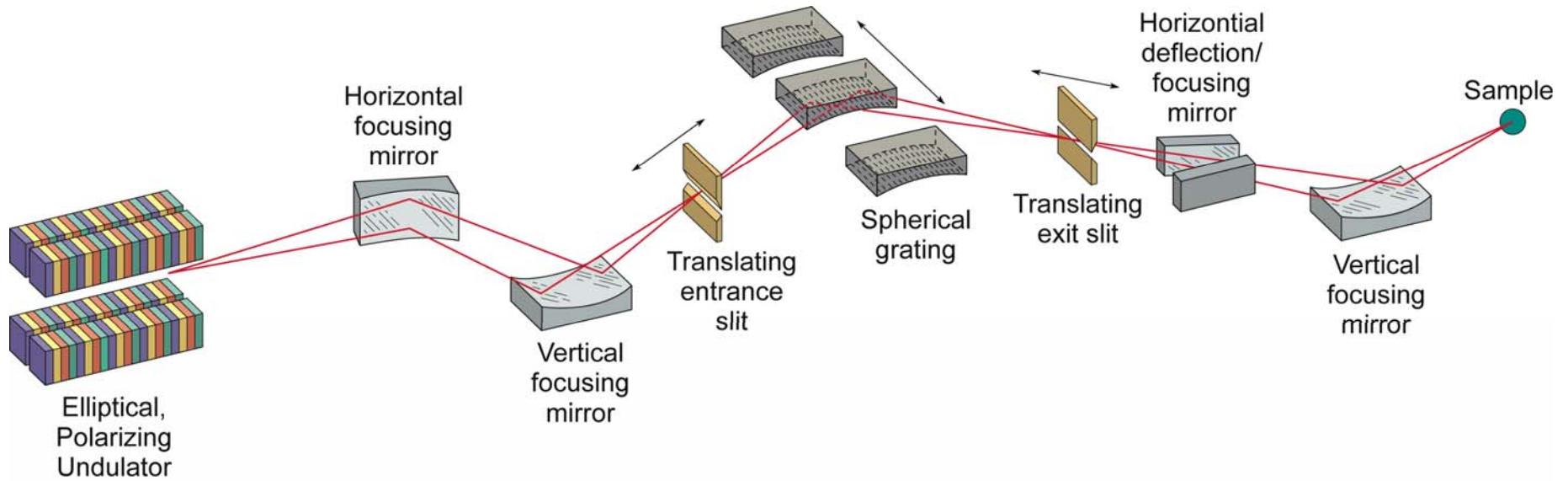


Courtesy of James Underwood (EUV Technology Inc.)

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High Spectral Resolution (meV) Beamline





Beamline 7.0 at Berkeley's Advanced Light Source

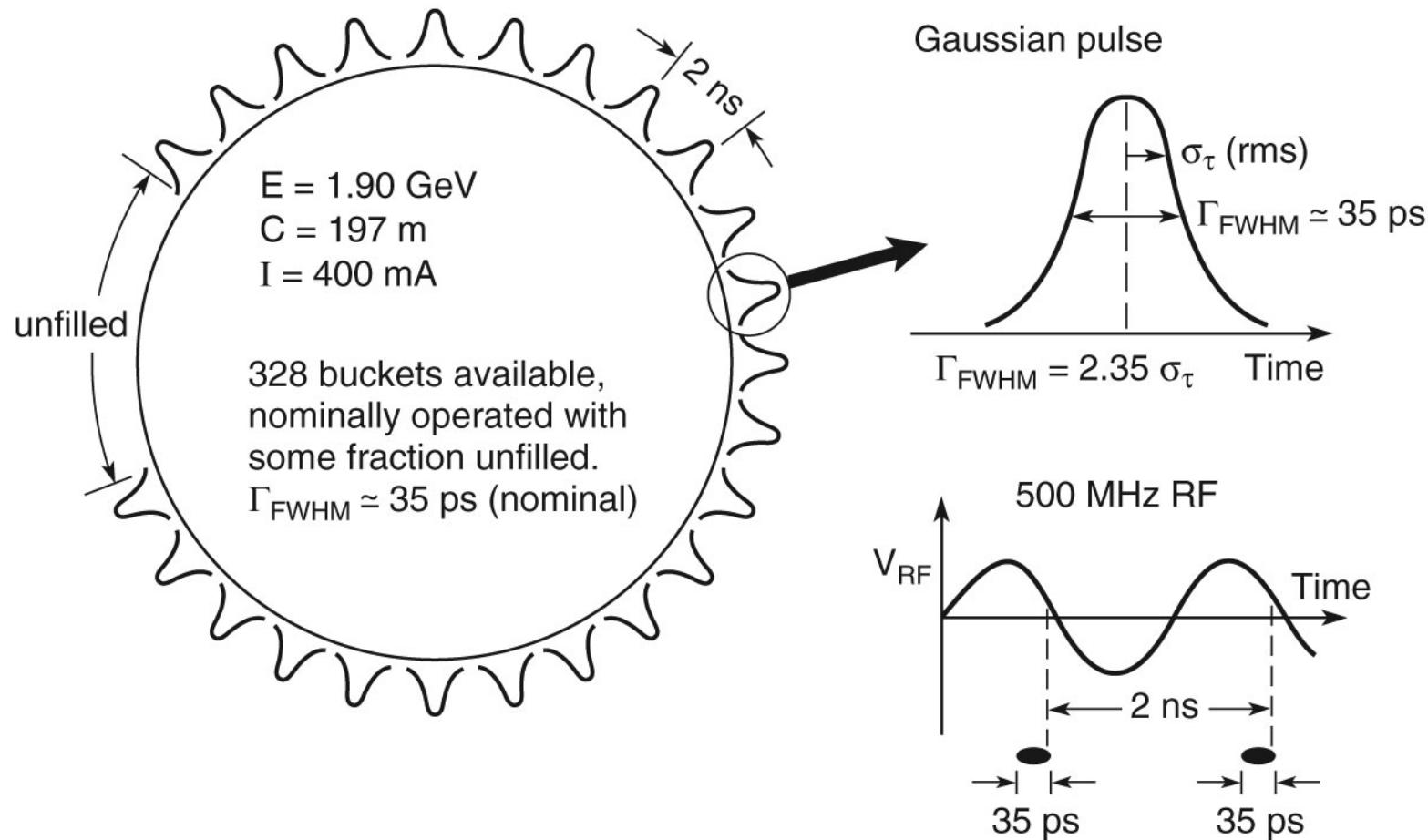




Time Structure of Synchrotron Radiation

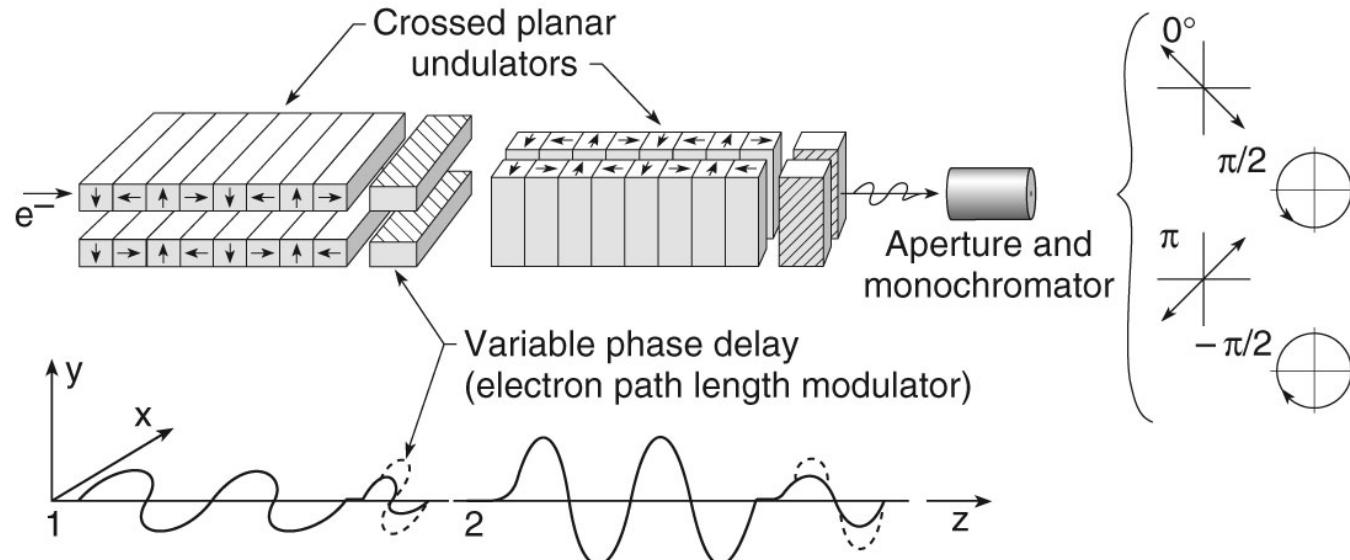


The axial electric field within the RF cavity, used to replenish lost (radiated) energy, forms a potential well “bucket” system that forces electrons into axial electron “bunches”. This leads to a time structure in the emitted radiation.

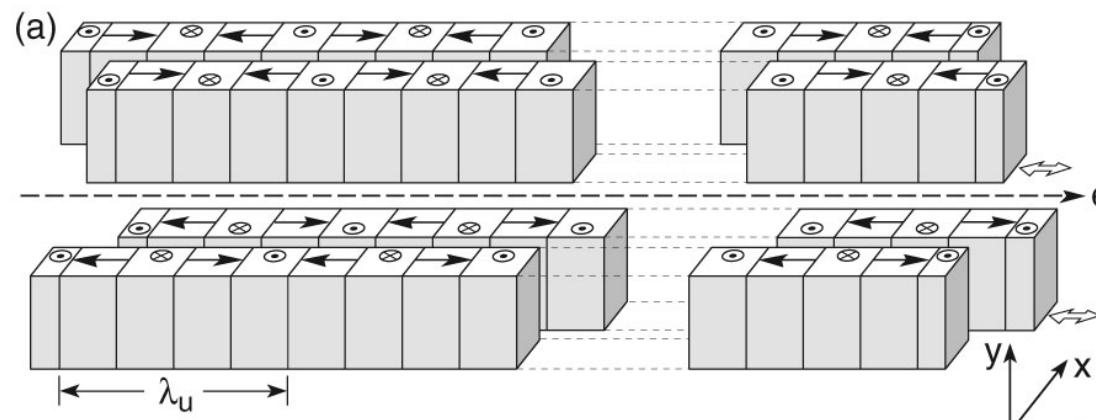




Variable Polarization Undulator Radiation



(Courtesy of Kwang-Je Kim)



(Following S. Sasaki)



What are the Relative Merits?



Bending magnet radiation

- Broad spectrum
- Good photon flux
- No heat load
- Less expensive
- Easier access

Wiggler radiation

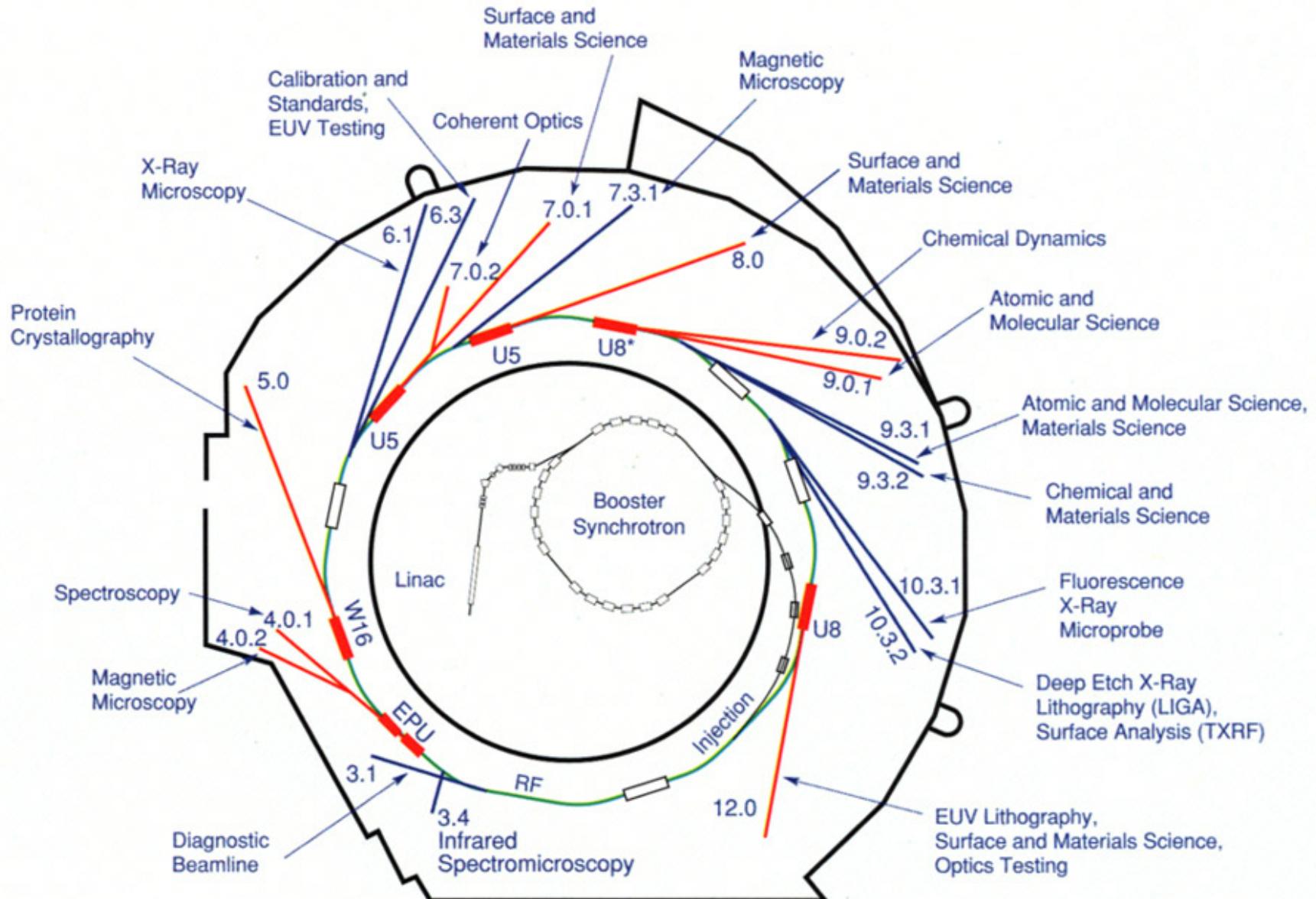
- Higher photon energies
- More photon flux
- Expensive magnet structure
- Expensive cooled optics
- Less access

Undulator radiation

- Brighter radiation
- Smaller spot size
- Partial coherence
- Expensive
- Less access



A Single Storage Ring Serves Many Scientific User Groups

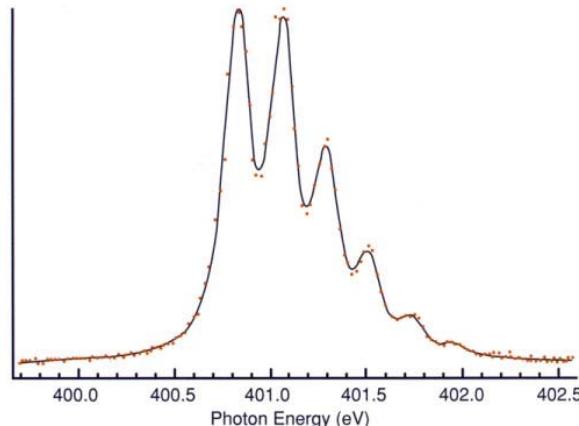




Some Sample Applications

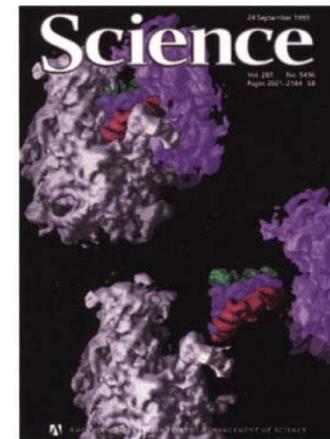


Absorption Fine Structure



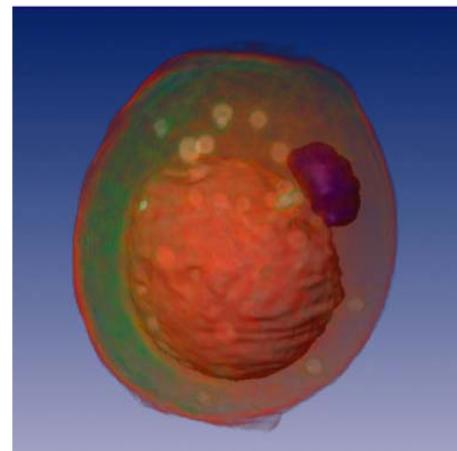
$1s \rightarrow \pi^*$ in N₂

Protein Crystallography



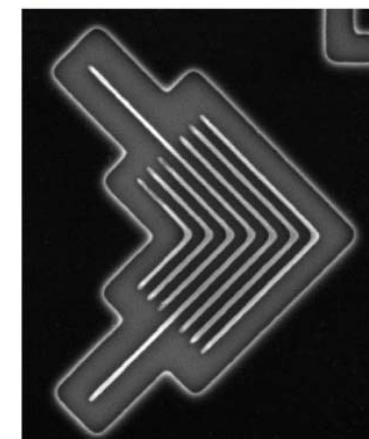
Ribosome structure

Soft X-ray Microscopy



Yeast cell

Printing Nanochip Patterns



39 nm elbow pattern



ALS Beamlines After 5 Years



Beamline	Source	Areas of Research	Techniques*	Energy Range	Monochromator	Available
3.1	Bend magnet	Diagnostic beamline		200-280 eV	None	Now
3.4	Bend magnet	Infrared spectromicroscopy	8, 9	0.05-1 eV	FTIR	1996
4.0.1	EPU elliptically polarizing undulator(s)	Spectroscopy	7, 12, 19, 20, 21	20-1800 eV	V-SGM	1996/7
4.0.2	EPU elliptically polarizing undulator(s)	Magnetic microscopy	6, 10, 12, 17	100-1600 eV	SGM	1996/7
5.0	W16 wiggler	Protein crystallography	22	4-13 keV	Double Crystal	1996
6.1.2	Bend magnet	High-resolution zone-plate microscopy	23, 24, 28	250-600 eV	Zone Plate Linear	Now
6.3.1	Bend magnet	Calibration and standards, EUV optics testing	11, 15, 24	500 eV-4 keV	Double Crystal	1995
6.3.2	Bend magnet	Calibration and standards, EUV optics testing	11, 15, 24, 25	50-1000 eV	VLS-PGM	1995
7.0.1	U5 undulator	Surface and materials science, spectromicroscopy	6, 7, 11, 14, 17, 19, 20, 21, 24	60-1000 eV	SGM	Now
7.0.2	U5 undulator	Coherent optics experiments	1, 27	70-650 eV	None	1996
7.3.1	Bend magnet	Magnetic microscopy	10, 12, 14, 17, 24	260-1200 eV	SGM	1996
8.0	U5 undulator	Surface and materials science	7, 20, 21	70-1200 eV	SGM	Now
9.0.1	U8 undulator †	Atomic and molecular science, EUV lithography	2, 5, 7, 15, 16, 18, 25	20-310 eV	SGM	Now
9.0.2	U8 undulator †	Chemical dynamics	2, 3, 16, 18	5-30 eV	White Light Eagle	1995
9.3.1	Bend magnet	Atomic and molecular science, materials science	2, 7, 15, 16, 18, 20, 21, 25	700 eV-6 keV	Double Crystal	1995
9.3.2	Bend magnet	Chemical and materials science	5, 7, 12, 15, 19, 20, 21	30-1500 eV	SGM	Now
10.3.1	Bend magnet	Fluorescence x-ray microprobe	6, 13	3-12 keV	Multilayer	Now
10.3.2	Bend magnet	Deep-etch x-ray lithography (LIGA), surface analysis (TXRF)	4, 26	3-12 keV	White Light	Now
12.0	U8 undulator	EUV lithography, surface and materials science, optics development	5, 17, 24	60-320 eV	VLG-PGM	1996

† Will change to U10 in September 1995

*See key



ALS Beamlines After 5 Years (continued)



Key to Techniques

1	Coherent scattering	15	Photoabsorption spectroscopy (incl. EXAFS, NEXAFS, XANES)
2	Coincidence detection	16	Photodissociation
3	Crossed-molecular-beam dynamics studies	17	Photoelectron microscopy
4	Deep-etch x-ray lithography (LIGA)	18	Photoelectron spectroscopy (incl. ZEKE)
5	EUV interferometry	19	Angle-resolved valence spectroscopy
6	Fluorescence microscopy	20	Core-level spectroscopy (ESCA, XPS)
7	Fluorescence spectroscopy	21	Diffraction/holography
8	Infrared microscopy	22	Protein crystallography
9	Infrared spectroscopy	23	Soft x-ray interferometry
10	Magnetic microscopy	24	Spectromicroscopy
	Magnetic spectroscopy	25	Time-of-flight spectroscopy
11	Faraday rotation	26	Total reflection x-ray fluorescence
12	Magnetic circular dichroism	27	X-ray holography
13	Micro-diffraction	28	X-ray transmission microscopy
14	Photoabsorption microscopy		

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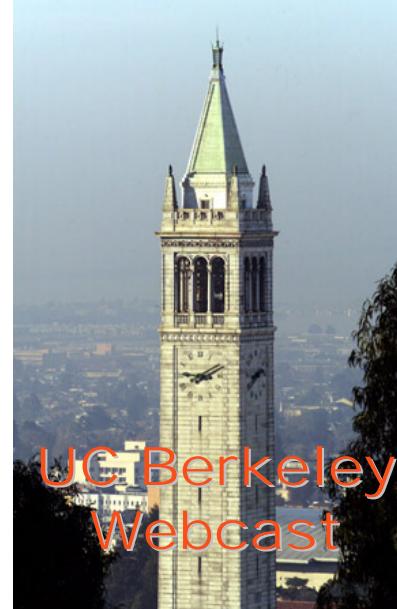
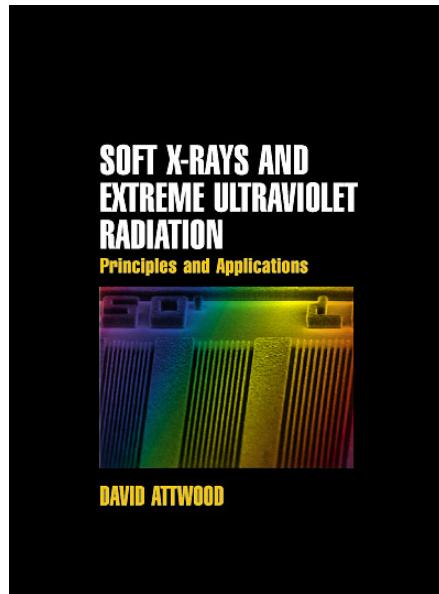
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- 1) D. Attwood, *Soft X-Rays and Extreme Ultraviolet Radiation* (Cambridge, UK, 2000).
- 2) P. Duke, *Synchrotron Radiation* (Oxford, UK, 2000).
- 3) J. Als-Nielsen and D. McMorrow, *Elements of Modern X-ray Physics* (Wiley, New York, 2001).
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- 5) A. Hofmann, *Synchrotron Radiation* (Cambridge, UK, 2004).
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